

DESCRIPTION

PLASMA DISPLAY PANEL WITH SUPERIOR LIGHT-EMITTING
CHARACTERISTICS, AND METHOD AND APPARATUS FOR
PRODUCING THE PLASMA DISPLAY PANEL

5 FIELD OF THE INVENTION

This invention relates to a plasma display panel used as a display for a color television receiver or the like, and also relates to a method of producing the plasma display panel.

10 BACKGROUND OF THE INVENTION

Recently, Plasma Display Panel (PDP) has received attention as a large-scale, thin, lightweight display for use in computers and televisions, and the demand for high-definition PDPs has also increased. Document EP0554172A1 discloses a
15 conventional, typical technique related to a construction and production method of PDP.

FIG. 29 is a sectional view showing a general AC-type PDP.

In the drawing, a front glass substrate 101 is covered
20 by a stack of display electrodes 102, a dielectric glass layer 103, and a dielectric protecting layer 104 in the order, where the dielectric protecting layer 104 is made of magnesium oxide

(MgO) (see, for example, Japanese Laid-Open Patent Application
No.5-342991.

baking the applied silver paste. The partition walls 107 are formed by applying the glass paste to the surface of the layers in stripes with a certain pitch, and baking the applied glass paste. The fluorescent substance layers 110 to 112 are formed
5 by applying fluorescent substance pastes of each color to the space between the partition walls, and baking the applied pastes at around 500°C to remove resin and other elements from the pastes. Japanese Laid-Open Patent Application No.2-08834 discloses a technique for forming a fluorescent substance film
10 by applying a fluorescent substance slurry then drying the applied slurry by high-temperature dry air.

After the fluorescent substances are baked, a sealing glass frit is applied to an outer region of the back glass substrate 105, then the applied sealing glass frit is baked at
15 around 350°C to remove resin and other elements from the applied sealing glass frit. (Frit Temporary Baking Process)

The front glass substrate 101 and the back glass substrate 105 are then put together so that the display electrodes 102 are perpendicular to the address electrodes 106,
20 the electrodes 102 facing the electrodes 106. The substrates are then bonded by heating them to a temperature (around 450°C) higher than the softening point of the sealing glass. (Bonding Process)

The bonded panel is heated to around 350°C while gases
25 are exhausted from inner space between the substrates (space

formed between the front and back substrates, where the fluorescent substances are in contact with the space) (Exhausting Process). After the exhausting process is

that the production cost is reduced.

The above first object is achieved by improving the chromaticity of light emitted from blue fluorescent substance layers. This is achieved by setting the chromaticity coordinate
5 y (the CIE color specification) of light to 0.07 or less or the peak wavelength of a spectrum of light to 453nm or less when vacuum ultraviolet rays are radiated onto the blue cells to excite the blue fluorescent substances.

Such an improvement in the chromaticity of light
10 emitted from blue fluorescent substance layers as described above increases the color temperature of light (white balance) when the light is emitted from all the cells, and improves the color reproduction.

The above PDP having a superior chromaticity of light
15 emitted from blue fluorescent substance layers is produced by performing the bonding process while steam vapor is forced to exhaust from the inner space by, for example, circulating a dry gas in the inner space.

The above PDP is also produced by performing a
20 preparative heating step before the bonding process, where in the preparative heating step, a front panel and a back panel are heated in an atmosphere of dry gas while a space is opened between the sides of the panels facing each other. Alternatively, the above PDP is produced by performing a heating
25 step before the bonding process, where in the heating step, a

panel is heated while an MgO layer formed on the panel is in contact with a dry gas.

The above improvement is achieved by the production method of the present invention since it prevents blue
5 fluorescent substances from being degraded by heat by reducing the amount of water preserved in the inner space. In contrast, in a conventional PDP production method, the blue fluorescent substances are degraded by heat of water emitted into the inner space in the bonding process, resulting in degradation of the
10 light-emitting intensity and the chromaticity of emitted light.

The above PDP whose blue fluorescent substance layers emit light with a superior chromaticity is also produced by performing the bonding process, after a while heating the bonded
15 panels to a certain temperature while circulating a dry gas in the inner space, and starting an exhausting step.

With the above construction, even if the chromaticity of light emitted from the blue fluorescent substance layers is degraded by heat of the water in the bonding process, the
20 chromaticity is recovered since the water is removed from the inner space as the dry gas is circulated in the inner space while the bonded panels are heated to the certain temperature.

Here, the "dry gas" indicates a gas containing steam vapor with lower partial pressure than the typical partial
25 pressure. It is preferable to use an air processed to be dried

(dry air).

It is desirable that the partial pressure of the steam vapor in the dry gas atmosphere is set to 15Torr or less, more preferably to 10Torr or less, 5Torr or less, 1Torr or less, 5 0.1Torr or less. It is desirable that the dew-point temperature of the dry gas is set to 20°C or lower, more preferably to 10°C or lower, 0°C or lower, -20°C or lower, -40°C or lower.

The above PDP with improved chromaticity of light emitted from blue fluorescent substance layers is also 10 manufactured by a PDP production method in which: the front and

back panels are temporarily baked while a space is opened between their facing sides; the front and back panels are bonded while a dry gas is circulated in an inner space between the panels; or the front and back panels are bonded together after
5 preparatively heated while a space is opened between their facing sides.

The second object of the present invention is achieved by: a method in which after the front panel and the back panel are bonded together by a sealing material in between by
10 maintaining a bonding temperature, the exhausting process is started while the panels are not cooled from the bonding temperature to room temperature, and gases are exhausted from the inner space between the panels; or a method in which after the front panel and the back panel with a sealing material in
15 between are temporarily baked by maintaining a temporary bonding temperature, then the bonding process is started while the panels are not cooled from the temporary bonding temperature to room temperature.

In the actual manufacturing procedure, each of such
20 heating processes is performed using a heating furnace. Conventionally, the sealing material temporary baking process, the bonding process, and the exhausting process are separately performed, and the panels are cooled to room temperature at each interval between processes. With such a construction, it
25 requires a long time and consumes much energy for the panels to

be heated in each process. In contrast, in the present invention, these processes are performed without lowering the temperature to room temperature. This reduces the time and energy required for heating.

5 BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a sectional view of the main part of the AC-type discharge PDP of Embodiment 1.

FIG. 2 shows a PDP display apparatus composed of the PDP shown in FIG. 1 and an activating circuit connected to the
10 PDP.

FIG. 3 shows a belt-conveyor-type heating apparatus used in Embodiment 1.

FIG. 4 shows the construction of a heating-for-sealing apparatus used in Embodiment 1.

15 FIG. 5 shows measurement results of the relative light-emitting intensity of light emitted from the blue fluorescent substance when it is baked in air with different partial pressures of the steam vapor contained in the air.

FIG. 6 shows measurement results of the chromaticity
20 coordinate y of light emitted from the blue fluorescent substance when it is baked in air with different partial pressures of the steam vapor contained in the air.

FIGs. 7A to 7C show measurement results of the number of molecules in H_2O gas desorbed from the blue fluorescent

substance.

FIGs. 8 to 16 show specific examples of Embodiment 2 concerning: the position of the air vents at the outer regions of the back glass substrate; and the format in which the sealing
5 glass frit is applied.

FIGs. 17 and 18 shows the characteristic of how the effect of recovering the once-degraded light-emitting characteristics depends on the partial pressure of steam vapor, where the blue fluorescent substance layer is once degraded then
10 baked again in air.

FIG. 19 shows the construction of a bonding apparatus used in the bonding process of Embodiment 5.

FIG. 20 is a perspective diagram showing the inner construction of the heating furnace of the bonding apparatus
15 shown in FIG.19.

FIGs. 21A to 21C show operations of the bonding apparatus in the preparative heating process and the bonding process.

FIG. 22 shows the results of the experiment for
20 Embodiment 5 in which the amount of steam vapor released from the MgO layer is measured over time.

FIG. 23 shows a variation of the bonding apparatus in Embodiment 5.

FIG. 24A to 24C show operations performed with another
25 variation of the bonding apparatus in Embodiment 5.

FIG. 25 shows spectra of light emitted from only blue cells of the PDPs of Embodiment 5.

FIG. 26 is a CIE chromaticity diagram on which the color reproduction areas around blue color are shown in relation to the PDPs of Embodiment 5 and the comparative PDP.

FIGs. 27A, 27B, and 27C show operations performed in the temporary baking process through the exhausting process using the bonding apparatus of Embodiment 6.

FIG. 28 shows the temperature profile used in the temporary baking process, bonding process, and exhausting process in manufacturing the panels of Embodiment 6.

FIG. 29 is a sectional view showing a general AC-type PDP.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

15 <Embodiment 1>

FIG. 1 is a sectional view of the main part of the AC-type discharge PDP in the present embodiment. The figure shows a display area located at the center of the PDP.

The PDP includes: a front panel 10 which is made up of a front glass substrate 11 with display electrodes 12 (divided into scanning electrodes 12a and sustaining electrodes 12b), a dielectric layer 13, and a protecting layer 14 formed thereon; and a back panel 20 which is made up of a back glass substrate 21 with address electrodes 22 and a dielectric layer

23 formed thereon. The front panel 10 and the back panel 20 are arranged so that the display electrodes 12 and the address electrodes 22 face each other. The space between the front panel 10 and the back panel 20 is divided into a plurality of discharge spaces 30 by partition walls 24 formed in stripes. Each discharge space is filled with a discharge gas.

Fluorescent substance layers 25 are formed on the back panel 20 so that each discharge space 30 has a fluorescent substance layer of one color out of red, green, and blue and that the fluorescent substance layers are repeatedly arranged in the order of the colors.

In the panel, the display electrodes 12 and address electrodes 22 are respectively formed in stripes, the display electrodes 12 being perpendicular to the partition walls 24, and the address electrodes 22 being parallel to the partition walls 24. A cell having one color out of red, green, and blue is formed at each intersection of a display electrode 12 and an address electrode 22.

The address electrodes 22 are made of metal (e.g., silver or Cr-Cu-Cr). To keep the resistance of the display electrodes low and to secure a large discharge area in the cells, it is desirable that each display electrode 12 consists of a plurality of bus electrodes (made of silver or Cr-Cu-Cr) with a small width stacked on a transparent electrode with a large width made of a conductive metal oxide such as ITO, SnO_2 ,

and ZnO. However, the display electrodes 12 may be made of silver like the address electrodes 22.

5 The dielectric layer 13, being a layer composed of a dielectric material, covers the entire surface of one side of the front glass substrate 11 including the display electrodes 12. The dielectric layer is typically made of a lead low-melting-point glass, though it may be made of a bismuth low-melting-point glass or a stack of a lead low-melting-point glass and a bismuth low-melting-point glass.

10 The protecting layer 14, being made of magnesium oxide, is a thin layer covering the entire surface of the dielectric layer 13.

15 The dielectric layer 23 is similar to the dielectric layer 13, but is further mixed with TiO_2 grains so that the layer also functions as a visible-light reflecting layer.

The partition walls 24, being made of glass, are formed to project over the surface of the dielectric layer 23 of the back panel 20.

20 The following are the fluorescent substances used in the present embodiment:

blue fluorescent substance $\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}$

green fluorescent substance $\text{Zn}_2\text{SiO}_4:\text{Mn}$

red fluorescent substance $\text{Y}_2\text{O}_3:\text{Eu}$.

25 The composition of these fluorescent substances is basically the same as that of conventional materials used in

PDP. However, compared with the conventional ones, the fluorescent substances of the present embodiment emit more excellently colored light. This is because the fluorescent substances are degraded by the heat added in the manufacturing process. Here, the emission of the excellently colored light means that the chromaticity coordinate y of the light emitted from blue cells is small (i.e., the peak wavelength of the emitted blue light is short), and that the color reproduction range near the blue color is wide.

In typical, conventional PDPs, the chromaticity coordinate y (CIE color specification) of the light emitted from blue cells when only blue cells emit light is 0.085 or more (i.e., the peak wavelength of the spectrum of the emitted light is 456nm or more), and the color temperature in the white balance without color correction (a color temperature when light is emitted from all of the blue, red, and green cells to produce a white display) is about 6,000K.

As a technique for improving the color temperature in the white balance, a technique is known in which the width of only the blue cells (pitch of the partition walls) is set to a large value, and the area of the blue cells is set to a value larger than that of the red or green cells. However, to set the color temperature to 7,000K or higher in accordance with this technique, the area of the blue cells should be 1.3 times that of the red or green cells, or more.

In contrast, In the PDP of the present embodiment, the chromaticity coordinate y of the light emitted from blue cells when only blue cells emit light is 0.08 or less, and the peak wavelength of the spectrum of the emitted light is 455nm or less. Under these conditions, it is possible to increase the color temperature to 7,000K or more in the white balance without color correction. Also, depending on the conditions at the manufacturing process, it is possible to decrease the chromaticity coordinate y even further, or increase the color temperature to 10,000K or more in the white balance without color correction.

As stated above, as the chromaticity coordinate y of blue cells becomes small, the peak wavelength of the emitted blue light becomes short. This will be explained later in Embodiments 3 and 5.

Later embodiments will also explain: why the color reproduction area becomes large as the chromaticity coordinate y of blue cells becomes small; and how the chromaticity coordinate y of the light emitted from blue cells is related to the color temperature in the white balance without color correction.

In the present embodiment, on the assumption that the present PDP is used for a 40-inch high definition TV, the thickness of the dielectric layer 13 is set to around $20\mu\text{m}$, and the thickness of the protecting layer 14 to around $0.5\mu\text{m}$.

Also, the height of the partition walls 24 is set to 0.1mm to 0.15mm, the pitch of the partition walls to 0.15mm to 0.3mm, and the thickness of the fluorescent substance layers 25 to 5 μ m to 50 μ m. The discharge gas is Ne-Xe gas in which Xe constitutes 5 50% in volume. The charging pressure is set to 500Torr to 800Torr.

The PDP is activated by the following procedure. As shown in FIG. 2, a panel activating circuit 100 is connected to the PDP. An address discharge is produced by applying a certain 10 voltage to an area between the display electrodes 12a and the address electrodes 22 of the cells to illuminate. A sustaining discharge is then produced by applying a pulse voltage to an area between the display electrodes 12a and 12b. The cells emit ultraviolet rays as the discharge proceeds. The emitted 15 ultraviolet rays are converted to visible light by the fluorescent substance layers 31. Images are displayed on the PDP as the cells illuminate through the above-described procedure.

Procedure of Producing PDP

20 The following are description of the procedure by which the PDP with the above construction is produced.

Producing the Front Panel

The front panel 10 is produced by forming the display electrodes 12 on the front glass substrate 11, covering it with 25 the dielectric layer 13, then forming the protecting layer 14 on

the surface of dielectric layer 13.

The display electrodes 12 are produced by applying silver pastes to the surface of the front glass substrate 11 with the screen printing method, then baking the applied silver pastes. The dielectric layer 13 is formed by applying a lead glass material (e.g., a mixed material of 70% by weight of lead oxide (PbO), 15% by weight of boron oxide (B_2O_3), and 15% by weight of silicon oxide (SiO_2)), then baking the applied material. The protecting layer 14 consisting of magnesium oxide (MgO) is formed on the dielectric layer 13 with the vacuum vapor deposition method or the like.

Producing the Back Panel

The back panel 20 is produced by forming the address electrodes 22 on the back glass substrate 21, covering it with the dielectric layer 23 (visible-light reflecting layer), then forming the partition walls 30 on the surface of the dielectric layer 23.

The address electrodes 22 are produced by applying silver pastes to the surface of the back glass substrate 21 with the screen printing method, then baking the applied silver pastes. The dielectric layer 23 is formed by applying pastes including TiO_2 grains and dielectric glass grains to the surface of the address electrodes 22, then baking the applied pastes. The partition walls 30 are formed by repeatedly applying pastes including glass grains with a certain pitch with the screen

printing method, then baking the applied pastes.

After the back panel 20 is made, the fluorescent substance pastes of red, green, and blue are made and applied to the space between the partition walls with the screen printing
5 method. The fluorescent substance layers 25 are formed by baking the applied pastes in air as will be described later.

The fluorescent substance pastes of each color are produced by the following procedure.

The blue fluorescent substance ($\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}$) is
10 obtained through the following steps. First, the materials, barium carbonate (BaCO_3), magnesium carbonate (MgCO_3), and aluminum oxide ($\alpha\text{-Al}_2\text{O}_3$), are formulated into a mixture so that the ratio Ba:Mg:Al is 1:1:10 in the atoms. Next, a certain amount of europium oxide (Eu_2O_3) is added to the above mixture.
15 Then, a proper amount of flax (AlF_3 , BaCl_2) is mixed with this mixture in a ball mill. The obtained mixture is baked in a reducing atmosphere (H_2 , N_2) at 1400°C to 1650°C for a certain time period (e.g., 0.5 hours).

The red fluorescent substance ($\text{Y}_2\text{O}_3:\text{Eu}$) is obtained
20 through the following steps. First, a certain amount of europium oxide (Eu_2O_3) is added to yttrium hydroxide $\text{Y}_2(\text{OH})_3$. Then, a proper amount of flax is mixed with this mixture in a ball mill. The obtained mixture is baked in air at 1200°C to 1450°C for a certain time period (e.g., one hour).

25 The green fluorescent substance ($\text{Zn}_2\text{SiO}_4:\text{Mn}$) is

obtained through the following steps. First, the materials, zinc oxide (ZnO) and silicon oxide (SiO₂), are formulated into a mixture so that the ratio Zn:Si is 2:1 in the atoms. Next, a certain amount of manganese oxide (Mn₂O₃) is added to the above
5 mixture. Then, a proper amount of flax is mixed with this mixture in a ball mill. The obtained mixture is baked in air at 1200°C to 1350°C for a certain time period (e.g., 0.5 hours).

The fluorescent substances of each color produced as above are then crushed and sifted so that the grains for each
10 color having a certain particle size distribution are obtained. The fluorescent substance pastes for each color are obtained by mixing the grains with a binder and a solvent.

The fluorescent substance layers 25 can be formed with methods other than the screen printing. For example, the
15 fluorescent substance layers may be formed by allowing a moving nozzle to eject a fluorescent substance ink, or by making a sheet of photosensitive resin including a fluorescent substance, attaching the sheet to the surface of the back glass substrate 21 on a side including partition walls 24, performing a
20 photolithography patterning then developing the attached sheet to remove unnecessary parts of the attached sheet.

Bonding Front Panel and Back Panel, Vacuum Exhausting, and Charging Discharge Gas

Sealing glass layers are formed by applying a sealing
25 glass frit to one or both of the front panel 10 and the back

panel 20 which have been produced as above. The sealing glass layers are temporarily baked to remove resin and other elements from the glass frit, which will be detailed later. The front panel 10 and the back panel 20 are then put together with the display electrodes 12 and the address electrodes 22 facing each other and being perpendicular to each other. The front panel 10 and the back panel 20 are then heated so that they are bonded together with the softened sealing glass layers. This will be detailed later.

The bonded panels are baked (for three hours at 350°C) while air is exhausted from the space between the bonded panels to produce a vacuum. The PDP is then completed after the discharge gas with the above composition is charged into the space between the bonded panels at a certain pressure.

Details of Baking Fluorescent Substance, Temporarily Baking Sealing Glass Frit, and Bonding Front Panel and Back Panel

The processes of baking the fluorescent substances, temporarily baking the sealing glass frit, and bonding the front panel and back panel will be described in detail.

FIG. 3 shows a belt-conveyor-type heating apparatus which is used to bake the fluorescent substances and temporarily bake the frit.

The heating apparatus 40 includes a heating furnace 41 for heating the substrates, a carrier belt 42 for carrying the substrates inside the heating furnace 41, and a gas guiding

pipe 43 for guiding an atmospheric gas into the heating furnace 41. The heating furnace 41 inside is provided with a plurality of heaters (not shown in the drawings) along the heating belt.

The substrates are heated with an arbitrary temperature profile by adjusting the temperatures near the plurality of heaters placed along the belt between an entrance 44 and an exit 45. Also, the heating furnace can be filled with the atmospheric gas injected through the gas guiding pipe 43.

Dry air can be used as the atmospheric gas. The dry air is produced by: allowing air to pass through a gas dryer (not shown in the drawing) which cools the air to a low temperature (minus tens °C); and condensing the steam vapor in the cooled air. The amount (partial pressure) of the steam vapor in the cooled air is reduced through this process and a dry air is finally obtained.

To bake the fluorescent substances, the back glass substrate 21 with the fluorescent substance layers 25 formed thereon is baked in the heating apparatus 40 in the dry air (at the peak temperature 520°C for 10 minutes). As apparent from the above description, the degradation caused by the heat and the steam vapor in the atmosphere during the process of baking the fluorescent substances is reduced by baking the fluorescent substances in a dry gas.

The lower the partial pressure of the steam vapor in the dry air is, the greater the effect on reducing the

degradation of the fluorescent substances by heat is. As a result, it is desirable that the partial pressure of the steam vapor is 15Torr or less. The above effect becomes more remarkable as the partial pressure of the steam vapor is set to
5 a lower value like 10Torr or less, 5Torr or less, 1Torr or less, 0.1Torr or less.

There is a certain relationship between the partial pressure of the steam vapor and the dew-point temperature. As a result, the above description can be rewritten by replacing
10 the partial pressure of the steam vapor with the dew-point temperature. That is, the lower the dew-point temperature is set to, the greater the effect on reducing the degradation of the fluorescent substances by heat is. It is therefore desirable that the dew-point temperature of the dry gas is set
15 to 20°C or lower. The above effect becomes more remarkable as the dew-point temperature of the dry gas is set to a lower value like 0°C or lower, -20°C or lower, -40°C or lower.

To temporarily bake the sealing glass frit, the front glass substrate 11 or the back glass substrate 21 with the
20 sealing glass layers formed thereon is baked in the heating apparatus 40 in the dry air (at the peak temperature 350°C for 30 minutes).

In this temporary baking process, as in the baking process, it is desirable that the partial pressure of the steam
25 vapor is 15Torr or less. Also, the effect is more remarkable as

the partial pressure of the steam vapor is set to a lower value like 10Torr or less, 5Torr or less, 1Torr or less, 0.1Torr or less. In other words, it is desirable that the dew-point temperature of the dry gas is set to 20°C or lower, and even
5 more desirable for the temperature to be set to a lower value like 0°C or lower, -20°C or lower, -40°C or lower.

FIG. 4 shows the construction of a heating-for-sealing apparatus.

A heating-for-sealing apparatus 50 includes a heating
10 furnace 51 for heating the substrates (in the present embodiment, the front panel 10 and the back panel 20), a pipe 52a for guiding an atmospheric gas from outside of the heating furnace 51 into the space between the front panel 10 and the back panel 20, and a pipe 52b for letting out the atmospheric
15 gas to the outside the heating furnace 51 from the space between the front panel 10 and the back panel 20. The pipe 52a is connected to a gas supply source 53 which supplies the dry air as the atmospheric gas. The pipe 52b is connected to a vacuum pump 54. Adjusting valves 55a and 55b are respectively attached
20 to the pipes 52a and 52b to adjust the flow rate of the gas passing through the pipes.

The front panel and back panel are bonded together as described below using the heating-for-sealing apparatus 50 with the above construction.

25 The back panel is provided with air vents 21a and 21b

at the outer regions which surround the display region. Glass pipes 26a and 26b are respectively attached to the air vents 21a and 21b. Please note that the partition walls and fluorescent substances that should be on the back panel 20 are omitted in
5 FIG. 4.

The front panel 10 and the back panel 20 are positioned properly with the sealing glass layers in between, then put into the heating furnace 51. In doing so, it is preferable that the positioned front panel 10 and the back panel
10 20 are tightened with clamps or the like to prevent shifts.

The air is exhausted from the space between the panels using the vacuum pump 54 to produce a vacuum there. The dry air is then sent to the space through the pipe 52a at a certain flow rate without using the vacuum pump 54. The dry air is exhausted
15 from the pipe 52b. That means the dry air flows through the space between the panels.

The front panel 10 and the back panel 20 are then heated (at the peak temperature 450°C for 30 minutes) while the dry air is flown through the space between the panels. In this
20 process, the front panel 10 and the back panel 20 are bonded together with the softened sealing glass layers 15.

After the bonding is complete, one of the glass pipes 26a and 26b is plugged up, and the vacuum pump is connected to the other glass pipe. The heating-for-sealing apparatus is used
25 in the vacuum exhausting process, the next process. In the

discharge gas charging process, a cylinder containing the discharge gas is connected to the other glass pipe, and the discharge gas is charged into the space between the panels operating an exhausting apparatus.

5 Effects of the Method Shown in the Present Embodiment

The method shown in the present embodiment of bonding the front and back panels has unique effects, which will be described below.

10 In general, gases like steam vapor are held by adsorption on the surface of the front panel and back panel. The adsorbed gases are released when the panels are heated.

In conventional methods, in the bonding process after the temporary baking process, the front panel and the back panel are first put together at room temperature, then they are heated
15 to be bonded together. In the bonding process, the gases held by adsorption on the surface of the front panel and back panel are released. Though a certain amount of the gases are released in the temporary baking process, gases are newly held by adsorption when the panels are laid in the air to room
20 temperature before the bonding process begins, and the gases are released in the bonding process. The released gases are confined in the small space between the panels. It is known by measurement that the partial pressure of the steam vapor in the space at this stage is typically 20Torr or more.

25 When this happens, the fluorescent substance layers

25 contacting the space are tend to be degraded by the heat and the gases confined in the space (among the gases, especially by the steam vapor released from the protecting layer 14). The degradation of the flourescent substance layers causes the
5 light-emitting intensity of the layers to decrease (especially the blue flourescent substance layer).

On the other hand, according to the method shown in the present embodiment, the degradation is reduced since the dry air is flown through the space when the panels are heated and
10 the steam vapor is exhausted from the space to the outside.

In this bonding process, like the flourescent substance baking process, it is desirable that the partial pressure of the steam vapor is 15Torr or less. Also, the degradation of the flourescent substance is reduced more as the
15 partial pressure of the steam vapor is set to a lower value like 10Torr or less, 5Torr or less, 1Torr or less, 0.1Torr or less. In other words, it is desirable that the dew-point temperature of the dry air is set to 20°C or lower, and even more desirable for the temperature to be set to a lower value like 0°C or
20 lower, -20°C or lower, -40°C or lower.

Study of Partial Pressure of Steam Vapor in Atmospheric Gas

It was confirmed by the experiments that the degradation of the blue flourescent substance due to heating can be prevented by reducing the partial pressure of the steam vapor
25 in the atmospheric gas.

FIGs. 5 and 6 respectively show the relative light-emitting intensity and the chromaticity coordinate y of the light emitted from the blue fluorescent substance ($\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}$). These values were measured after the blue fluorescent substance was baked in the air by changing the partial pressure of the steam vapor variously. The blue fluorescent substance was baked with the peak temperature 450°C maintained for 20 minutes.

The relative light-emitting intensity values shown in FIG. 5 are relative values when the light-emitting intensity of the blue fluorescent substance measured before it is baked is set to 100 as the standard value.

For obtaining the light-emitting intensity, first the emission spectrum of the fluorescent substance layer is measured using a spectro-photometer, next the chromaticity coordinate y is calculated from the measured emission spectrum, then the light-emitting intensity is obtained from a formula (light-emitting intensity = luminance / chromaticity coordinate y) with the calculated chromaticity coordinate y and a luminance measured beforehand.

Note that the chromaticity coordinate y of the blue fluorescent substance before it was baked was 0.052.

It is found from the results shown in FIGs. 5 and 6 that there is no reduction of light-emitting intensity by heat and that there is no change in the chromaticity when the partial

pressure of the steam vapor is around 0Torr. However, it is noted that as the partial pressure of the steam vapor increases, the relative light-emitting intensity of the blue fluorescent substance decreases and the chromaticity coordinate y of the
5 blue fluorescent substance increases.

It has conventionally been thought that the light-emitting intensity reduces and the chromaticity coordinate y increases when the blue fluorescent substance ($\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}$) because activating agent Eu^{2+} ion is oxidized through heating
10 and converted into Eu^{3+} ion (S.Oshio, T.Matsuoka, S.Tanaka, and H.Kobayashi, Mechanism of Luminance Decrease in $\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}^{2+}$ Phosphor by Oxidation, J.Electrochem.Soc., Vol.145, No.11, November 1988, pp.3903-3907). However, considering from the fact that the chromaticity coordinate y of the above blue
15 fluorescent substance depends on the partial pressure of the steam vapor in the atmosphere, it is thought that the Eu^{2+} ion does not directly react with oxygen in the atmospheric gas (e.g., air), but that the steam vapor in the atmospheric gas accelerates the reaction related to the degradation.

20 For comparison, reduction of the light-emitting intensity and change in the chromaticity coordinate y of the blue fluorescent substance ($\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}$) were measured for various heating temperatures. The measurement results show tendencies that reduction of the light-emitting intensity
25 increases as the heating temperature becomes higher in the range

of 300°C to 600°C, and that reduction of the light-emitting intensity increases as the partial pressure of the steam vapor becomes higher in any heating temperatures. On the other hand, though the measurement results show the tendency that change in the chromaticity coordinate y increases as the partial pressure of the steam vapor becomes higher, the measurement results do not show the tendency that change in the chromaticity coordinate y depends on the heating temperature.

Also, the amount of steam vapor released when heated was measured for each material constituting the front glass substrate 11, display electrodes 12, dielectric layer 13, protecting layer 14, back glass substrate 21, address electrodes 22, dielectric layer 23 (visible-light reflecting layer), partition walls 24, and fluorescent substance layers 25. According to the measurement results, MgO which is the material of the protecting layer 14 among others releases the largest amount of steam vapor. It is assumed from the results that the degradation of the fluorescent substance layers 25 by heat during bonding layer is mainly caused by the steam vapor released from the protecting layer 14.

Variations of the Present Embodiment

In the present embodiment, a certain amount of dry air is flown into the inner space between the panels during the bonding process. However, exhausting air from the inner space to produce a vacuum and injection of dry air may be repeated

alternately. By this operation, the steam vapor can effectively exhausted from the inner space and the degradation of the flourescent substance layer by heat can be reduced.

Also, all of the flourescent substance layer baking
5 process, temporary baking process, and bonding process may not necessarily be performed in the atmospheric dry gas. It is possible to obtain the same effect by performing only one or two processes among these in the atmospheric dry gas.

10 In the present embodiment, dry air as the atmospheric gas is flown into the inner space between the panels during the bonding process. However, it is possible to obtain a certain effect by flowing an inert gas such as nitrogen which does not react with the flourescent substance layer and whose partial pressure of the steam vapor is low.

15 In the present embodiment, dry air is forcibly injected into the inner space between panels 10 and 20 through the glass pipe 26a in the bonding process. However, the panels 10 and 20 may be bonded together in the atmosphere of dry air using, for example, the heating apparatus 40 shown in FIG. 3.
20 In this case, a certain effect is also obtained since a small amount of dry gas flows into the inner space through the air vents 21a and 21b.

Though not described in the present embodiment, the water held by adsorption on the surface of the protecting layer
25 14 decreases in amount when the front panel 10 with the

protecting layer 14 formed on its surface is baked in the atmospheric dry gas. With this performance only, the degradation of the blue fluorescent substance layer is restricted to a certain extent. It is expected that the effect
5 further increases by combining this method of baking the front panel 10 with the manufacturing process of the present embodiment.

The PDP manufactured in accordance with the method of the present embodiment has an effect of decreasing abnormal
10 discharge during PDP activation since the fluorescent substance layers contains a small amount of water.

Example 1

<Table 1>

TABLE 1

PANEL CONSTRUCTION AND LIGHT-EMITTING CHARACTERISTICS

PANEL No.	PARTIAL PRESSURE OF STEAM VAPOR IN ATMOSPHERIC GAS(Torr)		PANEL LUMINANCE (cd/m^2)	COLOR TEMPERATURE WHEN LIGHT IS EMITTED FROM ALL CELLS ON PANEL (K)	PEAK INTENSITY RATIO OF SPECTRUM OF BLUE AND GREEN LIGHT (BLUE/GREEN)	PEAK NUMBER OF MOLECULES IN H ₂ O GAS DESORBED FROM BLUE FLUORESCENT SUBSTANCE AT 200 °C OR MORE WITH TDS ANALYSIS	AXIS LENGTH RATIO OF BLUE FLUORESCENT SUBSTANCE CRYSTAL (c-AXIS/a-AXIS)
	BAKING FLUORESCENT SUBSTANCE	TEMPORARILY BAKING SEALING GLASS FRIT					
1	12.0	12.0	495	7100	0.80	1.0×10^{16}	4.02180
2	8.0	8.0	520	7500	0.88	7.9×10^{15}	4.02177
3	3.0	3.0	540	8400	1.02	5.3×10^{15}	4.02172
4	0.0	0.0	550	9000	1.10	2.2×10^{15}	4.02164
5	20.0	20.0	470	6300	0.76	2.6×10^{16}	4.02208

In Table 1, the panels 1 to 4 are PDPs manufactured based on the present embodiment. The panels 1 to 4 have been manufactured in different partial pressures of the steam vapor in the dry air flown during the fluorescent substance layer baking process, frit temporary baking process, and bonding process, the partial pressures of the steam vapor being in the range of 0Torr to 12Torr.

The panel 5 is a PDP manufactured for comparison. The panel 5 was manufactured in non-dry air (partial pressure of the steam vapor is 20Torr) through the fluorescent substance layer baking process, frit temporary baking process, and bonding process.

In each of the PDPs 1 to 5, the thickness of the fluorescent substance layer is $30\mu\text{m}$, and the discharge gas, Ne(95%)-Xe(5%), was charged with the charging pressure 500Torr.

Light Emitting Characteristics Test and the Results

For each of the panels (PDPs) 1 to 5, the panel luminance and the color temperature in the white balance without color correction (a panel luminance and a color temperature when light is emitted from all of the blue, red, and green cells to produce a white display), and the ratio of the peak intensity of the spectrum of light emitted from the blue cells to that of the green cells were measured as the light emitting characteristics.

The results of this test are shown in Table 1.

Each of the manufactured PDPs was disassembled and vacuum ultraviolet rays (central wavelength is 146nm) were radiated onto the blue fluorescent substance layers of the back panel using a krypton excimer lamp. The color temperature when light was emitted from all of the blue, red, and green cells, and the ratio of the peak intensity of the spectrum of light emitted from the blue cells to that of the green cells were then measured. The results were the same as the above ones since no color filter or the like was used in the manufactured front panel.

The blue fluorescent substances were then taken out from the panel. The number of molecules contained in one gram of H₂O gas desorbed from the blue fluorescent substances was measured using the TDS (Thermal Desorption) analysis method. Also, the ratio of c-axis length to a-axis length of the blue fluorescent substance crystal was measured by the X-ray analysis.

The above measurement was carried out as follows using an infrared-heating type TDS analysis apparatus made by ULVAC JAPAN Ltd.

Each test sample of fluorescent substance contained in a tantalum plate was housed in a preparative-exhausting chamber and gas was exhausted from the chamber to the order of 10^{-4} Pa. The test sample was then housed in a measuring

chamber, and gas was exhausted from the chamber to the order of 10^{-7} Pa. The number of H_2O molecules (mass number 18) desorbed from the fluorescent substance was measured in a scan mode at measurement intervals of 15 seconds while the test sample was
5 heated using an infrared heater from room temperature to $1,100^{\circ}C$ at heating rate $10^{\circ}C/min$. FIGs. 7A, 7B, and 7C show the test results for the blue fluorescent substances taken out from the panels 2, 4, and 5, respectively.

As observed from the drawing, the number of H_2O
10 molecules desorbed from the blue fluorescent substance has peaks at around $100^{\circ}C$ to $200^{\circ}C$ and at around $400^{\circ}C$ to $600^{\circ}C$. It is considered that the peak at around $100^{\circ}C$ to $200^{\circ}C$ is due to desorption of the physical adsorption gas, and the peak at around $400^{\circ}C$ to $600^{\circ}C$ is due to desorption of the chemical
15 adsorption gas.

Table 1 shows the peak value of the number of H_2O molecules desorbed at $200^{\circ}C$ or higher, namely H_2O molecules desorbed at around $400^{\circ}C$ to $600^{\circ}C$, and the ratio of c-axis length to a-axis length of the blue fluorescent substance crystal.

20 Study

By studying the results shown in Table 1, it is noted that the panels 1 to 4 of the present embodiment are superior to the panel 5 (comparative example) in the light emitting characteristics. That is, the panels 1 to 4 have higher panel
25 luminance and color temperatures.

In the panels 1 to 4, the light emitting characteristics increase in the order of the panel 1, 2, 3, 4.

It is found from this result that the light emitting characteristics (panel luminance and color temperature) become superior as the partial pressure of the steam vapor is lower in the fluorescent substance layer baking process, frit temporary baking process, and bonding process.

The reason for the above phenomenon is considered that when the partial pressure of the steam vapor is reduced, the degradation of the blue fluorescent substance layer ($\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}$) is prevented and the chromaticity coordinate y value becomes small.

In case of the panels of the present embodiment, the peak number of molecules contained in one gram of H_2O gas desorbed from the blue fluorescent substances at 200°C or higher is 1×10^{16} or less, and the ratio of c-axis length to a-axis length of the blue fluorescent substance crystal is 4.0218 or less. In contrast, the corresponding values of the comparative panel are both greater than the above values.

<Embodiment 2>

The PDP of the present embodiment has the same construction as that of Embodiment 1.

The manufacturing method of the PDP is also the same as Embodiment 1 except: the position of the air vents at the outer regions of the back glass substrate 21; and the format in

which the sealing glass frit is applied. During the bonding process, the fluorescent substance layer degrades by heat worse than during the fluorescent substance layer baking process and the frit temporary baking process since in the bonding process, the gas including the steam vapor being generated from the protecting layer, fluorescent substance layer, and sealing glass of the front panel is confined to each small inner space partitioned by the partition walls when heated. Considering this, in the present embodiment, it is arranged that the dry air injected into the inner space can flow steadily through the space between partition walls in the bonding process and that the gas generated in the space between partition walls is effectively exhausted. This increases the effect of preventing the degradation of the fluorescent substance layer by heat.

FIGs. 8 to 16 show specific embodiments concerning: the position of the air vents at the outer regions of the back glass substrate 21; and the format in which the sealing glass frit is applied. Note that though the back panel 20 is provided with the partition walls 24 in stripes over the whole image display area in reality, FIGs. 8 to 16 show only several columns of partition walls 24 for each of the sides, omitting the center part.

As shown in these figures, a frame-shaped sealing glass area 60 (an area on which the sealing glass layer 15 is formed) is allotted at the outer region of the back glass

substrate 21. The sealing glass area 60 is composed of: a pair of vertical sealing areas 61 extending along the outermost partition wall 24; and a pair of horizontal sealing areas 62 extending perpendicular to the partition walls (in the direction
5 of the width of the partition walls).

When panels are bonded together, dry air flows through gaps 65 between partition walls 24.

The characteristics of the present examples will be described with reference to the drawings.

10 As shown in FIGs. 8 to 12, air vents 21a and 21b are formed at diagonal positions inside the sealing glass area 60. When panels are bonded together, dry air guided through the air vent 21a, as shown in FIG. 4, passes through the gap 63a between the partition wall edge 24a and horizontal sealing area 62, is
15 divided into the gaps 65 between the partition walls 24. The dry air then passes through the gaps 65, passes through the gap 63b between the partition wall edge 24b and horizontal sealing area 62, and is exhausted from the air vent 21b.

In the example shown in FIG. 8, each of the gaps 63a
20 and 63b has greater width than each of the gaps 64a and 64b between the vertical sealing area 61 and the adjacent partition wall 24 (so that $D1, D2 > d1, d2$ is satisfied, where $D1, D2, d1,$ and $d2$ respectively represent the minimum widths of the gaps 63a, 63b, 64a, and 64b).

25 With such a construction, for the dry air supplied

through air vent 21a, the resistance to the gas flow in the gaps 65 between the partition walls 24 becomes smaller than that in the gaps 64a and 64b. As a result, a greater amount of dry air passes through gaps 63a and 63b than gaps 64a and 64b, resulting
5 in steady separation of the dry air into the gaps 65 and steady flow of the dry gas in the gaps 65.

With the above arrangement, the gas generated in each gap 65 is effectively exhausted, which enhances the effect of preventing the degradation of the fluorescent substance later in
10 the bonding process.

It can also be said that the greater values the minimum widths D1 and D2 of the gaps 63a and 63b are set to than the minimum widths d1 and d2 of the gaps 64a and 64b, such as two times or three times the values, the smaller the resistance
15 to the gas flow in the gaps 65 between the partition walls 24 becomes and the dry air flows through each gap 65 more steadily, further enlarging the effects.

In the example shown in FIG. 9, the center part of the vertical sealing area 61 is connected to the adjacent partition
20 wall 24. Therefore, the minimum widths d1 and d2 of the gaps 64a and 64b are each 0 around the center. In this case, the dry air flows through each gap 65 even more steadily since the dry air does not flow through the gaps 64a and 64b.

In the examples shown in FIGS. 10 to 16, a flow
25 preventing wall 70 is formed inside the sealing glass area 60 so

that they are in intimate contact. The flow preventing wall 70 is composed of: a pair of vertical walls 71 extending along the vertical sealing areas 61; and a pair of horizontal walls 72 extending along the horizontal sealing areas 62. The air vents 21a and 21b are adjacent to the flow preventing wall 70 inside. Note that in the example shown in FIG. 12, only horizontal walls 72 are formed.

The flow preventing wall 70 is made of the same material, with the same shape as the partition walls 24. As a result, they can be manufactured in the same process.

The flow preventing wall 70 prevents the sealing glass of the sealing glass area 60 from flowing into the display area located at the center of the panel when the sealing glass area 60 is softened by heat.

In the example shown in FIG. 10, as in the case shown in FIG. 8, each of the gaps 63a and 63b has greater width than each of the gaps 64a and 64b between the vertical sealing area 61 and the adjacent partition wall 24 (so that $D1, D2 > d1, d2$ is satisfied), providing the same effects as the case shown in FIG. 8.

In the example shown in FIG. 11, partitions 73a and 73b are formed respectively around the center of the gaps 64a and 64b between the vertical walls 71 and the adjacent partition walls 24. The minimum widths $d1$ and $d2$ of the gaps 64a and 64b are each 0 around the center, like the case shown in FIG. 9.

Therefore, this case also provides the same effects as the case shown in FIG. 9.

In the example shown in FIG. 12, the center part of the vertical sealing area 61 is connected to the adjacent partition wall 24. The minimum widths d_1 and d_2 of the gaps 64a and 64b are each 0 around the center, like the case shown in FIG. 9. Therefore, this case also provides the same effects as the case shown in FIG. 9.

In the example shown in FIG. 13, the air vents 21a and 21b are formed at the center of the gaps 64a and 64b between the vertical walls 71 and the adjacent partition walls 24, not at diagonal positions. In addition, partitions 73a and 73b are formed respectively at the edges of gaps 64a and 64b. Therefore, this case provides the same effects as the case shown in FIG. 11.

In the example shown in FIG. 14, two air vents 21a as inlets of gas and two air vents 21b as outlets of gas are formed, and a central partition wall 27 among the partition walls 24 is extended to connect to the horizontal walls 72 at both ends. Otherwise, the panel is almost the same as that shown in FIG. 11. In this case, dry air flows in each of the areas separated by the central partition wall 27. However, since each of the gaps 63a and 63b has greater width than each of the gaps 64a and 64b, this case also provides the same effects as the case shown in FIG. 11. Further, in the example

shown in FIG. 14, it is possible to adjust the flow rate of the dry air for each of the areas separated by the central partition wall 27.

Variations of the Present Embodiment

5 In the present embodiment, as in Embodiment 1, it is desirable that the partial pressure of the steam vapor is 15Torr or less (or the dew-point temperature of the dry air is 20°C or lower), and the same effect can be obtained by flowing, instead of the dry air, an inert gas such as nitrogen which does not
10 react with the flourescent substance layer and whose partial pressure of the steam vapor is low.

 The present embodiment describes the case in which partition walls are formed on the back panel. However, partition walls may be formed on the front panel in the same
15 way, gaining the same effects.

Example 2

<Table 2>

TABLE 2

PANEL LIGHT-EMITTING CHARACTERISTICS					
PANEL No.	PANEL LUMINANCE (cd/m ²)	COLOR TEMPERATURE WHEN LIGHT IS EMITTED FROM ALL CELLS ON PANEL (k)	PEAK INTENSITY RATIO OF SPECTRUM OF BLUE AND GREEN LIGHT (BLUE/GREEN)	PEAK NUMBER OF MOLECULES IN H ₂ O GAS DESORBED FROM BLUE FLUORESCENT SUBSTANCE AT 200 °C OR MORE WITH TDS ANALYSIS	AXIS LENGTH RATIO OF BLUE FLUORESCENT SUBSTANCE CRYSTAL (c-AXIS/a-AXIS)
6	540	8400	0.94	6.3×10^{15}	4.02175
7	500	7200	0.83	8.8×10^{15}	4.02177
8	470	6300	0.76	2.6×10^{16}	4.02208

The panel 6 is a PDP manufactured based on FIG. 10 of the present embodiment in which the partial pressure of the steam vapor in the dry air flown during the bonding process is set to 2Torr (the dew-point temperature of the dry air is set to -10).

The panel 7 is a PDP manufactured partially based on FIG. 15 of the present embodiment in which each of the gaps 63a and 63b has less width than each of the gaps 64a and 64b between the vertical sealing area 61 and the adjacent partition wall 24 (so that $D1, D2 < d1, d2$ is satisfied). Otherwise, the panel is manufactured based on FIG. 10. When the panel 7 is manufactured, panels are bonded together in the same conditions as the panel 6.

The panel 8 is a PDP manufactured for comparison. The panel 8 has one air vent 21a on the back panel 20, as shown in FIG. 16. During the bonding process, the front panel 10 and the back panel 20 were heated to bond together without flowing the dry air after they were put together.

The panels 6 to 8 were manufactured under the same conditions except the bonding process. The panels 6 to 8 have the same panel construction except the air vents and flow preventing walls. In each of the PDPs 6 to 8, the thickness of the fluorescent substance layer is $20\mu\text{m}$, and the discharge gas, Ne(95%)-Xe(5%), was charged with the charging pressure 500Torr.

Test for Light Emitting Characteristics

For each of the PDPs 6 to 8, the panel luminance and the color temperature in the white balance without color correction, and the ratio of the peak intensity of the spectrum of light emitted from the blue cells to that of the green cells were measured as the light emitting characteristics.

The results of this test are shown in Table 2.

Each of the manufactured PDPs was disassembled and vacuum ultraviolet rays were radiated onto the blue fluorescent substance layers of the back panel using a krypton excimer lamp. The color temperature when light was emitted from all of the blue, red, and green cells, and the ratio of the peak intensity of the spectrum of light emitted from the blue cells to that of the green cells were then measured. The results were the same as the above ones.

The blue fluorescent substances were then taken out from the panel. The number of molecules contained in one gram of H_2O gas desorbed from the blue fluorescent substances was measured using the TDS analysis method. Also, the ratio of c-axis length to a-axis length of the blue fluorescent substance crystal was measured by the X-ray analysis. The results are also shown in Table 2.

Study

By studying the results shown in Table 2, it is noted that the panel 6 of the present embodiment shows the best light

emitting characteristics among the three panels. The light emitting characteristics of the panel 6 are better than those of the panel 7. This is considered to be achieved for the following reasons: during the bonding process of the panel 6,
5 the dry air steadily flow through the gap between partition walls and the generated gas is effectively exhausted, while during the bonding process of the panel 7, almost all the dry air guided into the inside through the air vent 21a is exhausted to the outside through the air vent 21b after passing through
10 the gaps 63a and 63b; and in the case of panel 7, since a small amount of the dry gas flows through the gap 65 between the partition walls, the gas generated in the gap 65 is not effectively exhausted.

The light emitting characteristics of the panel 8 are
15 inferior to the others. This is also considered to be caused because the gas generated in the gap 65 is not effectively exhausted since a small amount of the dry gas flows through the gap 65 between the partition walls.

The PDPs in the present example are manufactured based
20 on FIG. 10. However, it has been confirmed that PDPs manufactured based on FIGs. 10 to 16 show similarly excellent light-emitting characteristics.

<Embodiment 3>

The PDP of the present embodiment has the same
25 construction as that of Embodiment 1.

The manufacturing method of the PDP is also the same as Embodiment 1 except: when the front panel 10 and the back panel 20 are bonded together in the bonding process, the panels are heated while the dry air is flown by adjusting the pressure of the inner space to be lower than atmospheric pressure.

In the present embodiment, first the sealing glass frit is applied onto one or both of the front panel 10 and back panel 20. The applied sealing glass frit is baked temporarily. The panels 10 and 20 are then put together and placed in the heating furnace 51 of the heating-for-sealing apparatus 50. Pipes 52a and 52b are respectively connected to the glass pipes 26a and 26b. The pressure of the inner space between panels is reduced by exhausting air from the space through the pipe 52b using the vacuum pump 54. At the same time, the dry air is supplied from the gas supply source 53 into the inner space through the pipe 52a at a certain flow rate. In doing so, adjusting valves 55a and 55b are adjusted to keep the pressure of the inner space lower than atmospheric pressure.

As described above, as the panels 10 and 20 are heated for 30 minutes at the sealing temperature (peak temperature is 450°C) while supplying the dry air into the inner space between panels under a reduced pressure, the sealing glass layer 15 is softened and the panels 10 and 20 are bonded together by the softened sealing glass.

The bonded panels are baked (for three hours at 350°C)

while air is exhausted from the inner space between the panels to produce a vacuum. The discharge gas with the above composition is then charged into the space at a certain pressure to complete the PDP.

5 Effects of the Present Embodiment

During the bonding process of the present embodiment, the panels are bonded together while dry gas is flown into the inner space between panels, as in Embodiment 1. Therefore, as described above, the degradation of the fluorescent substance caused by contacting with the steam vapor is restricted.

It is desirable, as in Embodiment 1, that the partial pressure of the steam vapor in the dry air is 15Torr or less. The effect of restricting the degradation becomes more remarkable as the partial pressure of the steam vapor is set to a lower value like 10Torr or less, 5Torr or less, 1Torr or less, 0.1Torr or less. It is desirable that the dew-point temperature of the dry gas is set to 20°C or lower, more desirably, to a lower value like 0°C or lower, -20°C or lower, -40°C or lower.

Further, in the present embodiment, the steam vapor generated in the inner space is more effectively exhausted to the outside than in Embodiment 1 since the panels are bonded together while the pressure of the inner space is kept to be lower than atmospheric pressure. The bonded panels 10 and 20 are in intimate contact since the inner space between panels does not expand during the bonding process since dry air is

supplied into the space while the pressure of the inner space is kept to be lower than atmospheric pressure.

The lower the pressure of the inner space is, the more easily the partial pressure of the steam vapor is adjusted to be low. This is desirable in terms of bonding the panels to be in intimate contact. Therefore, it is desirable to set the pressure of the inner space between panels to 500Torr or lower, more desirably to 300Torr or lower.

On the other hand, when the dry gas is supplied to the inner space between panels whose pressure is extremely low, the partial pressure of oxygen in the atmospheric gas becomes low. For this reason, oxide fluorescent substances such as $\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}$, $\text{Zn}_2\text{SiO}_4:\text{Mn}$, and $(\text{Y}_2\text{O}_3:\text{Eu})$ which are often used for PDPs cause defects like oxygen defects when heated in the atmosphere of non oxygen. This causes the light-emitting efficiency to be likely to decrease. Accordingly, from this point of view, it is desirable to set the pressure of the inner space to 300Torr or higher.

Variations of the Present Embodiment

In the present embodiment, dry air is supplied as the atmospheric gas into the inner space between the panels in the bonding process. However, the same effect can be obtained by flowing, instead of the dry air, an inert gas such as nitrogen which does not react with the fluorescent substance layer and whose partial pressure of the steam vapor is low. It should be

noted here that it is desirable to supply an atmospheric gas including oxygen in terms of restricting the degradation of the luminance.

In the present embodiment, the pressure of the inner space is reduced even when the temperature is too low to soften the sealing glass. In this case, however, gas may be flown into the inner space from the heating furnace 51 through gaps between the front panel 10 and back panel 20. As a result, it is desirable to supply or charge dry air to the heating furnace 51.

Alternatively, to prevent gas from flowing from the heating furnace 51 to the inner space between panels, the pressure of the inner space may be kept near atmospheric pressure by not exhausting the dry gas from the inner space when the temperature is still low and the sealing glass has not been softened, then the dry gas may be forcibly exhausted from the inner space after the temperature rises to a certain degree or more to reduce the pressure of the inner space to be lower than atmospheric pressure. In this case, it is desirable that the temperature at which the dry gas is forcibly exhausted is set to a degree at which the sealing glass begins to be softened, or higher. In this respect, it is preferable that the temperature at which the dry gas is forcibly exhausted is set to 300°C or higher, more preferably to 350°C or higher, and even more preferably to 400°C or higher.

The present embodiment describes the case in which during the bonding process, the panels 10 and 20 are heated while supplying the dry air into the inner space under a reduced pressure. However, the process of baking the fluorescent substances or temporarily baking the sealing glass frit may be performed in the atmosphere in which dry air is supplied under a reduced pressure. This provides a similar effect.

The application of the panel structure described in Embodiment 2 to the present embodiment produces further effects.

Example 3

<Table 3>

TABLE 3. PANEL BONDING CONDITIONS AND LIGHT-EMITTING CHARACTERISTICS

PANEL No.	DRY GAS TYPE	PARTIAL PRESSURE OF STEAM VAPOR IN DRY GAS (Torr)	PRESSURE IN SPACE BETWEEN PANELS (Torr)	TEMPERATURE FOR REDUCING TO BE LOWER THAN ATMOSPHERIC PRESSURE (°C)	RELATIVE LIGHT-EMITTING INTENSITY OF BLUE LIGHT	CHROMATICITY COORDINATE Y OF BLUE LIGHT	PEAK WAVELENGTH OF BLUE LIGHT (nm)	COLOR TEMPERATURE IN WHITE BALANCE (K)	PEAK INTENSITY RATIO OF SPECTRUM OF BLUE AND GREEN LIGHT (BLUE/GREEN)	PEAK NUMBER OF MOLECULES IN H ₂ O GAS DESORBED FROM BLUE FLUORESCENT SUBSTANCE AT 200 °C OR MORE WITH TDS ANALYSIS	AXIS LENGTH RATIO OF BLUE FLUORESCENT SUBSTANCE CRYSTAL (c-AXIS/a-AXIS)
11	AIR	12	500	370	108	0.075	455	7100	0.82	1.0×10 ¹⁶	4.02180
12	AIR	8	500	370	115	0.068	454	7600	0.88	7.9×10 ¹⁵	4.02177
13	AIR	3	500	370	120	0.063	453	7900	0.91	7.1×10 ¹⁵	4.02176
14	AIR	0	500	370	125	0.058	451	8700	0.96	5.9×10 ¹⁵	4.02174
15	AIR	0	300	370	120	0.058	451	8600	0.96	5.9×10 ¹⁵	4.02174
16	AIR	0	100	370	113	0.058	451	8500	0.95	5.3×10 ¹⁵	4.02172
17	AIR	0	500	ROOM TEMPERATURE	121	0.062	452	8000	0.92	6.4×10 ¹⁵	4.02176
18	AIR	0	500	320	123	0.060	452	8200	0.93	6.0×10 ¹⁵	4.02175
19	AIR	0	500	420	127	0.056	450	9000	0.98	2.2×10 ¹⁵	4.02164
20	NITROGEN	0	500	370	105	0.058	451	8400	0.94	4.8×10 ¹⁵	4.02173
21	Ne-Xe(%)	0	500	370	105	0.058	451	8400	0.94	4.8×10 ¹⁵	4.02173
22	AIR	0	ATMOSPHERIC PRESSURE	-	125	0.058	451	8700	0.96	5.9×10 ¹⁵	4.02174
23	.	.	ATMOSPHERIC PRESSURE	-	100	0.090	458	5800	0.67	2.6×10 ¹⁶	4.02208

Table 3 shows various conditions in which panels are bonded for respective PDPs which includes PDPs based on the present embodiment and PDPs for comparison.

5 The panels 11 to 21 are PDPs manufactured based on the present embodiment. The panels 11 to 21 have been manufactured in different conditions of: the partial pressure of the steam vapor in the dry gas flown into the inner space between panels during the bonding process; the gas pressure in the inner space between panels; the temperature at which the pressure of the inner space starts to be reduced to be lower than atmospheric pressure; and the type of the dry gas.

10 The panel 22 is a PDP manufactured based on Embodiment 1 in which the dry gas is supplied to the inner space, but gas is not forcibly exhausted from the space during the bonding process.

15 The panel 23 is a PDP manufactured for comparison. The panel 23 was manufactured based on a conventional method without supplying the dry air to the inner space between panels.

20 In each of the PDPs 11 to 23, the thickness of the flourescent substance layer is $30\mu\text{m}$, and the discharge gas, Ne(95%)-Xe(5%), was charged with the charging pressure 500Torr.

Test for Light Emitting Characteristics

25 For each of the PDPs 11 to 23, the relative light-

emitting intensity of the emitted blue light, the chromaticity coordinate y of the emitted blue light, the peak wavelength of the emitted blue light, the color temperature in the white balance without color correction, and the ratio of the peak intensity of the spectrum of light emitted from the blue cells to that of the green cells were measured as the light emitting characteristics.

Of the above characteristics, the relative light-emitting intensity of blue light, the chromaticity coordinate y of blue light, and the color temperature in the white balance without color correction were measured with the same method as Embodiment 1. The peak wavelength of the emitted blue light was measured by illuminating only the blue cells and measuring the emission spectrum of the emitted blue light. The results of this test are shown in Table 3.

Note that the relative light-emitting intensity values for blue light shown in Table 3 are relative values when the measured light-emitting intensity of the panel 23, a comparative example, is set to 100 as the standard value.

Each of the manufactured PDPs was disassembled and vacuum ultraviolet rays were radiated onto the blue fluorescent substance layers of the back panel using a krypton excimer lamp. The chromaticity coordinate y of blue light, the color temperature when light was emitted from all of the blue, red, and green cells, and the ratio of the peak intensity of the

spectrum of light emitted from the blue cells to that of the green cells were then measured. The results were the same as the above ones.

5 The blue fluorescent substances were then taken out from the panel. The number of molecules contained in one gram of H_2O gas desorbed from the blue fluorescent substances was measured using the TDS analysis method. Also, the ratio of c-axis length to a-axis length of the blue fluorescent substance crystal was measured by the X-ray analysis. The results are
10 also shown in Table 3.

Study

By studying the results shown in Table 3, it is noted that the panels 11 to 21 of the present embodiment have light emitting characteristics superior to those of the comparative
15 example (panel 23) (with higher light-emitting intensity of blue light and higher color temperature in the white balance).

The panels 14 and 22 have the same values for the light emitting characteristics. This shows that the same effects (light emitting characteristics) are gained if the
20 partial pressure of the steam vapor in the dry air flowing in the inner space is the same, regardless whether the pressure of the inner space is equivalent to or lower than the atmospheric pressure.

However, among the samples of the panel 22, some
25 samples were observed to have gaps between the partition walls

and the front panel. This is considered to be because the inner space expanded a little due to the dry gas supplied during the bonding process.

By comparing the light-emitting characteristics of the panels 11 to 14, it is noted that the light-emitting intensity of blue light increases and the chromaticity coordinate y of the emitted blue light decreases in the order of the panel 11, 12, 13, 14. This shows that the light-emitting intensity of emitted blue light increases and the chromaticity coordinate y of the emitted blue light decreases as the partial pressure of the steam vapor in the dry air decreases. This is considered to be because the degradation of the blue fluorescent substance is prevented by reducing the partial pressure of the steam vapor.

By comparing the light-emitting characteristics of the panels 14 to 16, it is noted that the panels have the same values for the chromaticity coordinate y of the emitted blue light. This shows that the chromaticity coordinate y of the emitted blue light is not affected by the pressure of the inner space between panels. It is also noted that the relative light-emitting intensity for blue light decreases in the order of the panel 14, 15, 16. This shows that the light-emitting intensity of emitted blue light decreases as the partial pressure of oxygen in the atmospheric gas decreases and defects like oxygen defects are generated in the fluorescent substance.

By comparing the light-emitting characteristics of the

panels 14, 20, and 21, it is noted that the panels have the same values for the chromaticity coordinate y of the emitted blue light. This shows that the chromaticity coordinate y of the emitted blue light is not affected by the type of the dry gas
5 flown into the inner space between panels. It is also noted that the relative light-emitting intensity for blue light of the panels 20 and 21 is lower than that of the panel 14. This shows that the light-emitting intensity of emitted blue light decreases since defects like oxygen defects are generated in the
10 flourescent substance when a gas such as nitrogen or Ne(95%)-Xe(5%) that does not contain oxygen is used as the dry gas.

By comparing the light-emitting characteristics of the panels 14 and 17 to 19, it is noted that the light-emitting intensity of blue light increases and the chromaticity
15 coordinate y of the emitted blue light decreases in the order of the panel 17, 18, 14, 19. This shows that the light-emitting intensity of emitted blue light increases and the chromaticity coordinate y of the emitted blue light decreases as the temperature at which gas starts to be exhausted from the inner
20 space to reduce the pressure of the inner space to be lower than atmospheric pressure is set to a higher degree. This is considered to be because setting the exhaust start temperature to a higher degree prevents the atmospheric gas around the panel from flowing into the inner space between panels.

25 By focusing attention on the relationships between the

chromaticity coordinate y of the emitted blue light and the peak wavelength of the emitted blue light for each panel provided in Table 3, it is noted that the peak wavelength is shorter as the chromaticity coordinate y is smaller. This shows that they are
5 proportional to each other.

<Embodiment 4>

The PDP of the present embodiment has the same construction as that of Embodiment 1.

The manufacturing method of the PDP is the same as
10 conventional methods up to the bonding process (i.e., during the bonding process, the front panel 10 and the back panel 20 put together are heated without the supply of dry air into the inner space between the panels). However, in the exhausting process, panels are heated while dry gas is supplied into the inner space
15 between the panels (hereinafter, this process is also referred to as a dry gas process) before gas is exhausted to produce a vacuum (vacuum exhausting process). This restores the light-emitting characteristics of the blue fluorescent substance layer to the level before they are degraded through the bonding
20 process or earlier.

The following are description of the exhausting process of the present embodiment.

In the exhausting process of the present embodiment, the heating-for-sealing apparatus shown in FIG. 4 is used, and
25 FIG. 4 will be referred to in the description.

The glass pipes 26a and 26b are respectively attached to the air vents 21a and 21b of the back panel 20 in advance. Pipes 52a and 52b are respectively connected to the glass pipes 26a and 26b. Gas is exhausted from the inner space between panels through the pipe 52b using the vacuum pump 54 to temporarily evacuate the inner space. Dry air is then supplied to the inner space at a certain flow rate through the pipe 52a without using the vacuum pump 54. This allows the dry air to flow through the inner space between the panels 10 and 20. The dry air is exhausted to the outside through the pipe 52b.

The panels 10 and 20 are heated to a certain temperature while the dry air is supplied to the inner space.

The supply of the dry air is then stopped. After this, the air is exhausted from the inner space between panels using the vacuum pump 54 while keeping the temperature at a certain degree to exhaust the gas held by adsorption in the inner space.

The PDP is completed after the discharge gas is charged into the cells after the exhausting process.

20 Effects of the Present Embodiment

The exhausting process of the present embodiment has the effect of preventing the degradation of the fluorescent substance layer from occurring during the process.

The exhausting process also has the effect of restoring the light-emitting characteristics of fluorescent

substance layers (especially of the blue fluorescent substance layer) to the level before they are degraded through the earlier processes. The fluorescent substance layers (especially the blue fluorescent substance layer) are susceptible to degradation by heat during the fluorescent substance layer baking process, temporary baking process, and bonding process. The exhausting process of the present embodiment recovers the light-emitting characteristics of fluorescent substance layers if they have been degraded during the above processes.

10 The reason for the above effects is thought to be as follows.

When the panels bonded together during the bonding process are heated, gas (especially steam vapor) is released in the inner space between the panels. For example, when the bonded panels are left in air, water is held by adsorption in the inner space. Therefore, steam vapor is released in the space between panels when the panels in this state are heated. According to the exhausting process of the present embodiment, such steam vapor is effectively exhausted to the outside since dry gas is flown through the inner space while the panels are heated before the vacuum exhausting process is started. Accordingly, compared with conventional exhausting processes in which gas is simply exhausted without supplying dry gas, the fluorescent substance is less degraded by heat during the exhausting process of the present embodiment.

It is also thought that the light-emitting characteristics are recovered since the gas exhausting process using the dry gas causes a reverse reaction to the degradation by heat to occur.

5 As apparent from the above description, the present embodiment provides a practically great effect that the once-degraded light-emitting characteristics of the blue fluorescent substance can be recovered in the exhausting process, the last heat process.

10 To enhance the effect of recovering the once-degraded light-emitting characteristics of the blue fluorescent substance, it is desired that the following conditions are satisfied.

15 The higher the peak temperature (i.e., the higher of: the temperature at which panels are heated while dry gas is supplied; and the temperature at which gas is exhausted to produce a vacuum) in the exhausting process is, the greater the effect of recovering the once-degraded light-emitting characteristics.

20 To obtain the effect sufficiently, it is preferable to set the peak temperature to 300°C or higher, more preferably to higher degrees such as 360°C or higher, 380°C or higher, and 400°C or higher. However, the temperature should not be set to such a high degree as softens the sealing glass to flow.

25 It is also preferable that the temperature at which

panels are heated while dry gas is supplied is set to be higher than the temperature at which gas is exhausted to produce a vacuum. This is because when the temperatures are set reversely, the effect is reduced by the gas (especially steam vapor) released from the panels into the inner space during the vacuum exhausting process; and when the temperatures are set as described above, the effect is obtained since the gas is released less from the panels into the inner space during the vacuum exhausting process than the former case.

It is preferred that the partial pressure of the steam vapor in the supplied dry gas is set to as low a value as possible. This is because the effect of recovering the once-degraded light-emitting characteristics of the blue fluorescent substance increases as the partial pressure of the steam vapor in the dry gas becomes low, though compared to conventional vacuum exhausting processes, the effect is remarkable when the partial pressure of the steam vapor is 15Torr or lower.

The following experiment also shows that it is possible to recover the once-degraded light-emitting characteristics of the blue fluorescent substance.

FIGs. 17 and 18 shows the characteristic of how the effect of recovering the once-degraded light-emitting characteristics depends on the partial pressure of steam vapor, where the blue fluorescent substance layer ($\text{BaMgAl}_{10}\text{O}_{17}:\text{Eu}$) is once degraded then baked again in air. The measurement method

is shown below.

The blue fluorescent substance (chromaticity coordinate y is 0.052) was baked (for 20 minutes at peak temperature 450°C) in air whose partial pressure of steam vapor was 30Torr so that the blue fluorescent substance was degraded by heat. In the degraded blue fluorescent substance, the chromaticity coordinate y was 0.092, and the relative light-emitting intensity (a value when the light-emitting intensity of the blue fluorescent substance measured before it is baked is set to 100 as the standard value) was 85.

The degraded blue fluorescent substance was baked again at certain peak temperatures (350°C and 450°C , maintained for 30 minutes) in air with different partial pressures of steam vapor. The relative light-emitting intensity and the chromaticity coordinate y of the re-baked blue fluorescent substances were then measured.

FIG. 17 shows relationships between the partial pressure of steam vapor in air at the re-baking and the relative light-emitting intensity measured after the re-baking. FIG. 18 shows relationships between the partial pressure of steam vapor in air at the re-baking and the chromaticity coordinate y measured after the re-baking.

It is noted from FIGs. 17 and 18 that regardless of whether the re-baking temperature is 350°C or 450°C , the relative light-emitting intensity of blue light is high and the

chromaticity coordinate y of blue light is small when the partial pressure of steam vapor in air at the re-baking is in the range of 0Torr to 30Torr. This shows that even if the flourescent substance is baked in an atmosphere including much steam vapor and the light-emitting characteristics are degraded, the light-emitting characteristics are recovered when the flourescent substance is baked again in an atmosphere whose partial pressure of steam vapor is low. That is, the results show that the degradation of the blue flourescent substance by heat is a reversible reaction.

It is also noted from FIGs. 17 and 18 that the effect of recovering the once-degraded light-emitting characteristics increases as the partial pressure of steam vapor in air at the re-baking decreases or the re-baking temperature increases.

A similar measurement was conducted for various periods during which the peak temperature is maintained, though the measurement is not detailed here. The results show that the effect of recovering the once-degraded light-emitting characteristics increases as the period during which the peak temperature is maintained increases.

Variations of the Present Embodiment

In the present embodiment, dry air is used when panels are heated in the exhausting process. However, inert gas such as nitrogen or argon can be used instead of the dry air and the same effects can be obtained.

In the exhausting process of the present embodiment, panels are heated while dry air is supplied into the space between the panels before the vacuum exhausting starts. However, by setting the temperature during the vacuum exhausting process to a degree higher than the general degree (i.e., to 360°C or higher), the light-emitting characteristics of the fluorescent substance can be recovered to a certain extent by performing only the vacuum exhausting process. Also in this case, the higher the exhausting temperature is, the greater the effect of recovering the light-emitting characteristics is.

however, the exhausting process of the present embodiment has greater effect of recovering the light-emitting characteristics than the above variation. It is thought this is because in case of the above variation, a sufficient amount of steam vapor is not exhausted to outside the panels in the vacuum exhausting process since the inner space between panels is small.

It is expected that application of the panel construction described in Embodiment 2 to the present embodiment will enhance the effect of exhausting gas when panels are heated while dry gas is supplied.

Example 4

<Table 4>

TABLE 4

PANEL VACUUM EXHAUST CONDITIONS AND LIGHT-EMITTING CHARACTERISTICS(BLUE LIGHT)					
PANEL No.	HEATING TEMPERATURE DURING DRY AIR SUPPLY(°C) (MAINTAINED FOR 30 MINUTES)	HEATING TEMPERATURE DURING VACUUM EXHAUST (°C) (MAINTAINED FOR TWO HOURS)	PARTIAL PRESSURE OF STEAM VAPOR IN DRY AIR(Torr)	RELATIVE LIGHT-EMITTING INTENSITY OF BLUE LIGHT	CHROMATICITY COORDINATE Y OF BLUE LIGHT
21	350	350	2	107	0.062
22	360	350	2	110	00.61
23	390	350	2	118	0.056
24	410	350	2	125	0.053
25	410	410	2	121	0.056
26	350	410	2	105	0.065
27	410	350	12	112	0.070
28	410	350	8	116	0.067
29	410	350	0	128	0.052
30	—	360	—	103	0.085
31	—	390	—	107	0.081
32	—	410	—	110	0.076
33	—	350	—	100	0.090

The panels 21 to 29 are PDPs manufactured based on the present embodiment. The panels 21 to 29 have been manufactured at different heating or exhausting temperatures when panels are heated while dry gas is supplied into the inner space. In this
5 process, a certain heating temperature was maintained for 30 minutes while dry gas was supplied into the inner space, then in the next vacuum exhausting process, a certain exhausting temperature was maintained for two hours.

10 The panels 30 to 32 are PDPs manufactured based on the variation of the present embodiment. The panels 30 to 32 have been manufactured without the dry gas process, performing the vacuum exhausting process at 360°C or higher.

15 The panel 33 is a PDP manufactured based on a conventional method. The panel 33 was manufactured without the dry gas process, performing the vacuum exhausting process at 350°C for two hours.

In each of the PDPs 21 to 33, the thickness of the fluorescent substance layer is 30μm, and the discharge gas, Ne(95%)-Xe(5%), was charged with the charging pressure
20 500Torr.

Test for Light Emitting Characteristics

For each of the PDPs 21 to 33, the relative light-emitting intensity of blue light and the chromaticity coordinate y of blue light were measured as the light emitting
25 characteristics.

<Test Results and Study>

The results of this test are shown in Table 4. Note that the relative light-emitting intensity values for blue light shown in Table 4 are relative values when the measured light-emitting intensity of the comparative panel 33 is set to 100 as the standard value.

As noted from Table 4, each of the panels 21 to 28 has higher light-emitting intensity and smaller chromaticity coordinate y than the panel 33. This shows that the light-emitting characteristics of PDPs are improved by adopting the exhausting process of the present embodiment when manufacturing PDPs.

By comparing the light-emitting characteristics of the panels 21 to 24, it is noted that the light-emitting characteristics are improved in the order of panels 21, 22, 23 and 24 (the light-emitting intensity increases and the chromaticity coordinate y decreases). This shows that the higher a degree the heating temperature of the dry gas process is set to, the greater the effect of recovering the light-emitting characteristics of the blue fluorescent substance layer is.

By comparing the light-emitting characteristics of the panels 24 to 26, it is noted that the light-emitting characteristics are improved in the order of panels 26, 25, and 24. This shows that the higher a degree the heating temperature

of the dry gas process is set to than the exhausting temperature of the vacuum exhausting process, the greater the effect of recovering the light-emitting characteristics of the blue fluorescent substance layer is.

5 By comparing the light-emitting characteristics of the panels 24, and 27 to 29, it is noted that the light-emitting characteristics are improved in the order of panels 27, 28, 24, and 29. This shows that the smaller a value the partial pressure of steam vapor of the dry gas process is set to, the
10 greater the effect of recovering the light-emitting characteristics of the blue fluorescent substance layer is.

Each of the panels 30 to 32 has higher light-emitting intensity and smaller chromaticity coordinate y than the panel 33. This shows that the light-emitting characteristics of PDPs
15 are improved by adopting the exhausting process that is the variation of the present embodiment in manufacturing PDPs.

Each of the panels 30 to 32 has lower light-emitting characteristics than the panel 21. This shows that the effect of recovering the light-emitting characteristics of the blue
20 fluorescent substance layer is greater when the dry gas process of the present embodiment is adopted.

<Embodiment 5>

The PDP of the present embodiment has the same construction as that of Embodiment 1.

25 The manufacturing method of the PDP of the present

embodiment is the same as Embodiment 1 up to the temporary baking process. However, in the bonding process, panels are preparatively heated while space is made between the facing sides of the panels, then the heated panels are put together and
5 bonded together.

In the PDP of the present embodiment, the chromaticity coordinate y of the light emitted from blue cells when light is emitted from only blue cells is 0.08 or less, the peak wavelength of the spectrum of the emitted light is 455nm or
10 less, and the color temperature is 7,000K or more in the white balance without color correction. Further, it is possible to increase the color temperature in the white balance without color correction to about 11,000K depending on the manufacturing conditions by setting the chromaticity coordinate y of blue
15 light to 0.06 or less.

Now, the bonding process of the present embodiment will be described in detail.

FIG. 19 shows the construction of a bonding apparatus used in the bonding process.

20 The bonding apparatus 80 includes a heating furnace 81 for heating the front panel 10 and the back panel 20, a gas supply valve 82 for adjusting the amount of atmospheric gas supplied into the heating furnace 81, a gas exhaust valve 83 for
adjusting the amount of the gas exhausted from the heating
25 furnace 81.

The inside of the heating furnace 81 can be heated to a high temperature by a heater (not illustrated). An atmospheric gas (e.g., dry air) can be supplied into the heating furnace 81 through the gas supply valve 82, the atmospheric gas forming the atmosphere in which the panels are heated. The gas can be exhausted from the heating furnace 81 through the gas exhaust valve 83 using a vacuum pump (not illustrated) to produce a vacuum in the heating furnace 81. The degree of vacuum in the heating furnace 81 can be adjusted with the gas supply valve 82 and the gas exhaust valve 83.

A dryer (not illustrated) is formed in the middle of the heating furnace 81 and an atmospheric gas supply source. The dryer cools the atmospheric gas (to minus several tens degree) to remove the water in the atmospheric gas by condensing water in the gas. The atmospheric gas is sent to the heating furnace 81 via the dryer so that the amount of steam vapor (partial pressure of steam vapor) in the atmospheric gas is reduced.

A base 84 is formed in the heating furnace 81. On the base 84, the front panel 10 and the back panel 20 are laid. Slide pins 85 for moving the back panel 20 to positions parallel to itself are formed on the base 84. Above the base 84, pressing mechanisms 86 for pressing the back panel 20 downwards are formed.

FIG. 20 is a perspective diagram showing the inner

construction of the heating furnace 81.

In FIGs. 19 and 20, the back panel 20 is placed so that the length of the partition walls is represented as a horizontal line.

5 As shown in FIGs. 19 and 20, the length of the back panel 20 is greater than that of the front panel 10, both edges of the back panel 20 extending off the front panel 10. Note that the extended parts of the back panel 20 are provided with leads which connect the address electrodes 22 to the activating
10 circuit. The slide pins 85 and the pressing mechanisms 86 are positioned at the four corners of the back panel 20, sandwiching the extended parts of the back panel 20 in between.

The four slide pins 85 protrude from the base 84 and can be simultaneously moved upwards and downwards by a pin
15 hoisting and lowering mechanism (not illustrated).

Each of the four pressing mechanisms 86 is composed of a cylindrical-shaped supporter 86a fixed on the ceiling of the heating furnace 81, a slide rod 86b which can move upwards and downwards inside the supporter 86a, and a spring 86c which
20 adds pressure on the slide rod 86b downwards inside the supporter 86a. With the pressure given to the slide rod 86b, the back panel 20 is pressed downwards by the slide rod 86b.

FIGs. 21A to 21C show operations of the bonding apparatus in the preparative heating process and the bonding
25 process.

The temporary baking, preparative heating, and bonding processes will be described with reference to FIGs. 21A to 21C.

Temporary Baking Process

5 A paste made of a sealing glass (glass frit) is applied to one of: the outer region of the front panel 10 on a side facing the back panel 20; the outer region of the back panel 20 on a side facing the front panel 10; and the outer region of the front panel 10 and the back panel 20 on sides that face each other. The panels with the paste are temporarily baked for 10 to 30 minutes at around 350°C to form the sealing glass layers 15. Note that in the drawing, the sealing glass layers 15 are formed on the front panel 10.

Preparative Heating Process

15 First, the front panel 10 and the back panel 20 are put together after positioned properly. The panels are then laid on the base 84 at a fixed position. The pressing mechanisms 86 are then set to press the back panel 20 (FIG. 21A).

20 The atmospheric gas (dry air) is then circulated in the heating furnace 81 (or, at the same time, gas is exhausted through the gas exhaust valve 83 to produce a vacuum) while the following operations are performed.

25 The slide pins 85 are hoisted to move the back panel 20 to a position parallel to itself (FIG. 21B). This broadens

the space between the front panel 10 and the back panel 20, and the fluorescent substance layers 25 on the back panel 20 are exposed to the large space in the heating furnace 81.

The heating furnace 81 in the above state is heated
5 to let the panels release gas. The preparative heating process ends when a preset temperature (e.g., 400°C) has been reached.

Bonding Process

The slide pins 85 are lowered to put the front and back panels together again. That is, the back panel 20 is reset
10 to its proper position on the front panel 10 (FIG. 21C).

When the inside of the heating furnace 81 has reached a certain bonding temperature (around 450°C) higher than the softening point of the sealing glass layers 15, the bonding temperature is maintained for 10 to 20 minutes. During this
15 period, the outer regions of the front panel 10 and the back panel 20 are bonded together by the softened sealing glass. Since the back panel 20 is pressed onto the front panel 10 by the pressing mechanisms 86 during this bonding period, the panels are bonded with high stability.

20 After the bonding is complete, the pressing mechanisms 86 are released and the bonded panels are removed.

The exhausting process is performed after the bonding process is performed as above.

In the present embodiment, as shown in FIGs. 19 and
25 20, an air vent 21a is formed on the outer region of the back

panel 20. The gas exhaust is performed using a vacuum pump (not illustrated) connected to a glass pipe 26 which is attached to the air vent 21a. After the exhausting process, the discharge gas is charged into the inner space between the panels through the glass pipe 26. The PDP is then complete after the air vent 21a is plugged and the glass pipe 26 is cut away.

Effects of the Manufacturing Method Shown in the Present Embodiment

The manufacturing method of the present embodiment has the following effects which are not obtained from the conventional methods.

As explained in Embodiment 1, with the conventional methods, the fluorescent substance layers 25 contacting the inner space between the panels are tend to be degraded by the heat and the gases confined in the space (among the gases, especially by the steam vapor released from the protecting layer 14). The degradation of the fluorescent substance layers causes the light-emitting intensity of the layers to decrease (especially the blue fluorescent substance layer).

According to the method shown in the present embodiment, though gases like steam vapor held by adsorption on the front and back panels are released during the preparative heating process, the gases are not confined in the inner space since the panels are separated with broad space in between. Further, since the panels are heated to be bonded together

immediately after the preparative heating, water and the like are not held by adsorption on the panels after the preparative heating. Therefore, less gas is released from the panels 10 and 20 during the bonding process, preventing the fluorescent substance layer 25 from degrading by heat.

Further, in the present embodiment, the preparative heating process through the bonding process are performed in the atmosphere in which dry air is circulated. Therefore, there is no degradation of the fluorescent substance layer 25 by heat and the steam vapor included in the atmospheric gas.

Another advantage of the present embodiment is that since the preparative heating process and the bonding process are consecutively performed in the same heating furnace 81, the processes can be performed speedily, consuming less energy.

Also, by using the bonding apparatus with the above construction, it is possible to bond the front panel 10 and the back panel 20 at a properly adjusted position.

Studies on Temperature in Preparative Heating and Timing with which Panels are Put together

It is considered to be desirable that the panels are heated to as high a temperature as possible in terms of preventing the fluorescent substance layer 25 from degrading by heat and the gases released from the panels when they are bonded (among the gases, especially by the steam vapor released from the protecting layer 14).

The following experiments were conducted to study the problem in detail.

The amount of steam vapor released from the MgO layer was measured using a TDS analysis apparatus over time while a glass substrate on which the MgO layer is formed as the front panel 10 is gradually heated at a constant heating speed.

FIG. 22 shows the results of the experiment, or the measured amount of released steam vapor at each heating temperature up to 700°C.

10 In FIG. 22, the first peak appears at around 200°C to 300°C, and the second peak at around 450°C to 500°C.

It is estimated from the results shown in FIG. 22 that a large amount of steam vapor is released at around 200°C to 300°C and around 450°C to 500°C when the protecting layer 14 is gradually heated.

15 Accordingly, to prevent the steam vapor released from the protecting layer 14 from being confined in the inner space when the panels are heated during the bonding process, it is considered that the separation of the panels should be maintained while they are heated at least until the temperature rises to around 200°C, preferably to around 300°C to 400°C.

Also, the release of gas from the panels will be almost completely prevented if the panels are bonded together after they are heated to a temperature higher than around 450°C while they are separated. In this case, the change of panels

over time after they are completed will also be prevented since the panels are bonded together with the fluorescent substance hardly degraded and with almost no chances that the steam vapor held by adsorption on the panels is gradually released during
5 discharging.

However, it is not preferable that this temperature exceeds 520°C since the fluorescent substance layer and the MgO protective layer are generally formed at the baking temperature of around 520°C . As a result, it is further preferable that the
10 panels are bonded together after they are heated to around 450°C to 520°C .

On the other hand, the sealing glass will flow out of the position if the panels are heated to a temperature exceeding the softening point of the sealing glass while they are
15 separated. This may inhibit the panels from being bonded with high stability.

From the view point of preventing the degradation of the fluorescent substance layer by the gases released from the panels and in terms of bonding the panels with high stability,
20 the following conclusions (1) to (3) are reached.

(1) It is desirable that the front and back panels are put together and bonded after heated to as high a temperature as possible under the softening point of the used sealing glass while the panels are separated from each other.

25 Accordingly, when, for example, a conventionally used

general sealing glass with softening point of around 400°C is used, to reduce the bad effect of released gases on the fluorescent substance as much as possible while maintaining the stability of the bonding, the best bonding procedure will be to
5 heat the front and back panels to near 400°C while separating them, then to put the panels together and heat them to a temperature exceeding the softening point to bond them together.

(2) Here, use of a sealing glass with a higher softening point
10 will increase the heating temperature and enhance the stability of bonding the panels. Accordingly, using such a high-softening point sealing glass to heat the front and back panels to near the softening point, then putting the panels together and heat them to a temperature exceeding the softening point to bond them
15 together will further reduce the bad effect of released gases on the fluorescent substance while maintaining the stability of bonding panels.

(3) On the other hand, it is possible to bond the panels with high stability even if they are heated, while they are
20 separated, to a high temperature exceeding the softening point of the sealing glass if an arrangement is made so that the sealing glass layer formed on the outer region of the front or back panel does not flow out of the position even if it is softened. For example, a partition may be formed between the
25 sealing glass application area and the display area at the outer

region of the front or back panel in order to prevent the softened sealing glass from flowing out to the display area.

Accordingly, when the front and back panels are heated to a high temperature exceeding the softening point of the sealing glass after making such an arrangement for preventing the softened sealing glass from flowing out to the display area and then the panels are put together and bonded together, the bad effect of the released gases on the fluorescent substance can be reduced, with the stability in bonding panels being kept.

In the above case, the front and back panels are bonded together directly at a high temperature without being put together first then being heated. As a result, release of gases from the panels after they are put together can almost completely be prevented. This enables the panels to be bonded together with almost no degradation of the fluorescent substance by heat.

Study on Atmospheric Gas and Pressure

It is desirable that a gas containing oxygen like air is used as the atmospheric gas circulated in the heating furnace 81 during the bonding process. This is because, as described in Embodiment 1, oxide fluorescent substances often used for PDPs tend to reduce the light-emitting characteristics when heated in the atmosphere of non oxygen.

A certain degree of effect can be gained when outside

air is supplied as the atmospheric gas at normal pressure. However, to enhance the effect of preventing the fluorescent substance from degrading, it is desirable to circulate dry gas like dry air in the heating furnace 81, or operate the heating
5 furnace 81 while exhausting gas to produce a vacuum.

The reason it is desirable to circulate dry gas is that there is no worrying that the fluorescent substance is degraded by heat and the steam vapor contained in the atmospheric gas. Also, it is desirable to exhaust gas from the
10 heating furnace 81 to produce a vacuum. This is because gases (steam vapor and the like) released from the panels 10 and 20 as they are heated are effectively exhausted to outside.

When dry gas is circulated as an atmospheric gas, the lower the partial pressure of steam vapor contained in the gas
15 is, the more the blue fluorescent substance layer is prevented from being degraded by heat (see FIGs. 5 and 6 for the experiment results of Embodiment 1). To obtain sufficient effect, it is desirable to set the partial pressure of the steam vapor to 15Torr or less. This effect becomes more remarkable as
20 the partial pressure of the steam vapor is set to a lower value like 10Torr or less, 5Torr or less, 1Torr or less, 0.1Torr or less.

Application of Sealing Glass

In the bonding process, the sealing glass is typically
25 applied to only one of the two panels (typically to the back

panel only) before the panels are put together.

Meanwhile, in the present embodiment, the back panel 20 is pressed onto the front panel 10 by the pressing mechanisms 86 in the bonding apparatus 80. In this case, it is difficult
5 to give such a strong pressure as is given by clamps.

In such a case, when the sealing glass is applied only to the back panel, there is a possibility that the panels are not completely bonded if the congeniality between the sealing glass and the front panel is not good in relation to adhesion.
10 This defect can be prevent if the sealing glass layer is formed on both the front and back panels. This will increase the manufacturing yield of PDPs.

It should be noted here that the above method of forming the sealing glass layer on both the front and back
15 panels is effective in increasing yields for the general bonding process in manufacturing PDPs.

Variations of Present Embodiment

In the present embodiment, the front panel 10 and the back panel 20 are put together after positioned properly before
20 they are heated. The slide pins 85 are then hoisted to move the back panel 20 upwards and separate the panels. However, the panels 10 and 20 may be separated from each other by other ways.

For example, FIG. 23 shows another way of lifting the
25 back panel 20. In the drawing, the front panel 10 is enclosed

with a frame 87, where the front panel 10 fits into the frame 87. The frame 87 can be moved upwards and downwards by rods 88 which are attached to the frame 87 and slide vertically. With such an arrangement, the back panel 20 laid on the frame 87 can
5 also be moved upwards and downwards to positions parallel to itself. That is, the back panel 20 is separated from the front panel 10 when the frame 87 is moved upwards, and the back panel 20 is put together with the front panel 10 when the frame 87 is moved downwards.

10 There is another difference between the two mechanisms. In the bonding apparatus 80, the back panel 20 is pressed onto the front panel 10 by the pressing mechanisms 86, while in the example shown in FIG. 23, a weight 89 is laid on the back panel 20 instead of the pressing mechanisms 86. In
15 this variation method, when the frame 87 is moved downwards to the bottom, the weight 89 presses the back panel 20 onto the front panel 10 by gravitation.

FIGs. 24A to 24C show operations performed during the bonding process in accordance with another variation method.

20 In the example shown in FIGs. 24A to 24C, the back panel 20 is partially separated from the front panel 10 and restored to the initial position.

On the base 84, as in the case shown in FIG. 20, four pins, or a pair of pins 85a and a pair of pins 85b are formed on
25 the base 84 corresponding to the four corners of the back panel

20. However, the pins 85a corresponding to one side (in FIGs. 24A to 24C, on the left-hand side) of the back panel 20, support the back panel 20 at their edges (e.g., the edge of the pin 85a formed in a spherical shape is fitted into a spherical pit
5 formed on the back panel 20), while the pins 85b corresponding to the other side (in FIGs. 24A to 24C, on the right-hand side) of the back panel 20 are movable upwards and downwards.

The front panel 10 and the back panel 20 are put together and laid on the base 84 as shown in FIG. 24A. The back
10 panel 20 is rotated about the edge of the pins 85a by moving the pins 85b upwards as shown in FIG. 24B. This partially separate the back panel 20 from the front panel 10. The back panel 20 is rotated in the reversed direction and restored to the initial position by moving the pins 85b downwards as shown in FIG. 24C.
15 That is, the panels 10 and 20 are in the same position as are adjusted properly at first.

The panels 10 and 20 are in contact at the side of pins 85a in the stage shown in FIG. 24B. However, gases released from panels are not confined in the inner space since
20 the other side of the panels are open.

Example 5

<Table 5>

TABLE 5.

PANEL No.	TEMPERATURE FOR PUTTING FRONT AND BACK PANELS TOGETHER(°C)	PEAK TEMPERATURE FOR BONDING PANELS(°C)	ATMOSPHERE DURING BONDING	PARTIAL PRESSURE OF STEAM VAPOR IN DRY AIR (Torr)	RELATIVE LIGHT-EMITTING INTENSITY OF BLUE LIGHT	CHROMATICITY COORDINATE Y OF BLUE LIGHT	COLOR TEMPERATURE IN WHITE BALANCE (K)	CHROMATICITY COORDINATE Y OF BLUE LIGHT WHEN BACK PANEL FLUORESCENT SUBSTANCE IS RADIATED BY EXCIMER LAMP	PEAK WAVE-LENGTH OF BLUE LIGHT (NM)	COLOR TEMPERATURE OF LIGHT WHEN BACK PANEL FLUORESCENT SUBSTANCES OF ALL COLORS ARE RADIATED BY EXCIMER LAMP (K)	PEAK INTENSITY RATIO OF SPECTRUM OF BLUE AND GREEN LIGHT (BLUE/GREEN)	PEAK NUMBER OF MOLECULES IN H ₂ O GAS DESORBED FROM BLUE FLUORESCENT SUBSTANCE AT 200 °C OR MORE WITH TDS ANALYSIS	AXIS LENGTH RATIO OF BLUE FLUORESCENT SUBSTANCE CRYSTAL (c-AXIS/a-AXIS)
41	250	450	DRY AIR	2	107	0.078	6700	0.075	455	6700	0.80	1.0×10^{16}	4.02180
42	350	450	DRY AIR	2	118	0.057	8600	0.054	451	8600	0.95	4.0×10^{15}	4.02172
43	400	450	DRY AIR	12	108	0.075	7100	0.073	459	7100	0.82	7.3×10^{15}	4.02178
44	400	450	DRY AIR	8	112	0.065	7800	0.063	452	7800	0.91	5.0×10^{15}	4.02174
45	400	450	DRY AIR	2	120	0.055	9000	0.054	450	9000	0.98	3.4×10^{15}	4.02168
46	400	450	DRY AIR	0	123	0.053	9800	0.052	449	9800	1.09	2.2×10^{15}	4.02164
47	400	450	VACUUM	-	120	0.053	9300	0.052	449	9300	1.03	1.3×10^{15}	4.02163
48	450	450	DRY AIR	2	125	0.052	10600	0.051	448	10600	1.15	1.9×10^{15}	4.02160
49	500	500	DRY AIR	2	125	0.052	10600	0.051	448	10600	1.15	1.9×10^{15}	4.02160
50	450	480	DRY AIR	2	126	0.052	11000	0.051	448	11000	1.19	1.3×10^{15}	4.02155
51	450	450	DRY AIR	2	125	0.052	10600	0.051	448	10600	1.15	1.9×10^{15}	4.02160
52	-	450	DRY AIR	2	100	0.090	5800	0.088	458	5800	0.67	2.6×10^{16}	4.02208

The panels 41 to 50 are PDPs manufactured based on the present embodiment. The panels 41 to 50 have been manufactured in different conditions during the bonding process. That is, the panels were heated in various types of atmospheric gases under various pressures, and they were put together at various temperatures with various timing.

Each panel had been temporarily baked at 350°C.

For the panels 41 to 46, 48 to 50, dry gases with different partial pressures of steam vapor in the range of 0Torr to 12Torr were used as the atmospheric gas. The panel 47 was heated while gas was exhausted to produce a vacuum.

For the panels 43 to 47, the panels were heated from the room temperature to 400°C (lower than the softening point of sealing glass), then the panels were put together. The panels were further heated to 450°C (higher than the softening point of sealing glass), the temperature was maintained for 10 minutes then decreased to 350°C, and gas was exhausted while the temperature of 350°C was maintained.

For the panels 41 and 42, the panels were bonded at lower temperatures of 250°C and 350°C, respectively.

For the panel 48, the panels were heated to 450°C, then put together at the temperature. For the panel 49, the panels were heated to 500°C (peak temperature), then put together at the temperature.

For the panel 50, the panels were heated to the peak

temperature of 480°C then decreased to 450°C, and the panels were put together and bonded at 450°C.

The panel 51 is a PDP manufactured based on a variation of Embodiment 5 shown in FIGs. 24A to 24C in which the
5 panels were heated to 450°C (peak temperature), then put together and bonded at the temperature.

The panel 52 is a comparative PDP manufactured by putting the panels together at room temperature then bonding them by heating them to 450°C in dry air at atmospheric
10 pressure.

Note that in each of the PDPs 41 to 52, the thickness of the fluorescent substance layer is 30μm, and the discharge gas, Ne(95%)-Xe(5%), was charged with the charging pressure 500Torr so that each has the same panel construction.

15 Test for Light Emitting Characteristics

For each of the PDPs 41 to 52, the relative light-emitting intensity of the emitted blue light, the chromaticity coordinate y of the emitted blue light, the peak wavelength of the emitted blue light, the panel luminance and the color
20 temperature in the white balance without color correction, and the ratio of the peak intensity of the spectrum of light emitted from the blue cells to that of the green cells were measured as the light emitting characteristics.

Each of the manufactured PDPs was disassembled and
25 vacuum ultraviolet rays (central wavelength is 146nm) were

radiated onto the blue fluorescent substance layers of the back panel using a krypton excimer lamp. The chromaticity coordinate y of blue light was then measured.

The results are shown in Table 5. Note that the
5 relative light-emitting intensity values for blue light shown in Table 5 are relative values when the measured light-emitting intensity of the panel 52, a comparative example, is set to 100 as the standard value.

Also, each of the manufactured PDPs was disassembled
10 and vacuum ultraviolet rays were radiated onto the blue fluorescent substance layers of the back panel using a krypton excimer lamp. The the color temperature when light was emitted from all of the blue, red, and green cells, and the ratio of the peak intensity of the spectrum of light emitted from the blue
15 cells to that of the green cells were then measured. The results were the same as the above ones.

FIG. 25 shows spectra of light emitted from only blue cells of the PDPs of panels 45, 50, and 52.

Though Table 5 does not show, the chromaticity
20 coordinate x and y of light emitted from the red and green cells of 41 to 53 were substantially the same: red (0.636, 0.350), green (0.251, 0.692). In the comparative PDP, the chromaticity coordinate x and y of light emitted from blue cells was (0.170, 0.090), and the peak wavelength was 458nm in the spectrum of the
25 emitted light.

The blue fluorescent substances were then taken out from the panel. The number of molecules contained in one gram of H_2O gas desorbed from the blue fluorescent substances was measured using the TDS analysis method. Also, the ratio of c-axis length to a-axis length of the blue fluorescent substance crystal was measured by the X-ray analysis. The results are also shown in Table 5.

Study

It is noted that the panels 41 to 51 have light emitting characteristics superior to those of the panel 52 (with higher light-emitting intensity of blue light and smaller chromaticity coordinate y). It is thought that this is because a smaller amount of gas is released in the inner space between panels after the panels are bonded in accordance with the present embodiment than in accordance with conventional methods.

In the PDP of panel 52, the chromaticity coordinate y of the light emitted from blue cells is 0.088 and the color temperature in the white balance without color correction is 5800K. In contrast, in panels 41 to 51, the values are respectively 0.08 or less and 6500K or more. Especially, it is noted that in panels 48 to 51 that have low chromaticity coordinate y of blue light, a high color temperature of around 11,000K has been achieved (in the white balance without color correction).

FIG. 26 is a CIE chromaticity diagram on which the color reproduction areas around blue color are shown in relation to the PDPs of the present embodiment and the comparative example.

5 In the drawing, the area (a) indicates the color reproduction area around blue color for a case (corresponding to panel 52) in which the chromaticity coordinate y of blue light is about 0.09 (the peak wavelength of spectrum of emitted light is 458nm), the area (b) indicates the color reproduction area
10 around blue color for a case (corresponding to panel 41) in which the chromaticity coordinate y of blue light is about 0.08 (the peak wavelength of spectrum of emitted light is 455nm), and the area (c) indicates the color reproduction area around blue color for a case (corresponding to panel 50) in which the
15 chromaticity coordinate y of blue light is about 0.052 (the peak wavelength of spectrum of emitted light is 448nm).

It is noted from the drawing that the color reproduction area around blue color expands in the order of area (a), (b), (c). This shows that it is possible to manufacture a
20 PDP in which the smaller the chromaticity coordinate y of blue light is (the shorter the peak wavelength of the spectrum of emitted light is), the broader the color reproduction area around blue color is.

By comparing the light-emitting characteristics of the
25 panels 41, 42, 45, and 48 (in each of which the partial pressure

of steam vapor in the dry gas is 2Torr), it is noted that the light-emitting characteristics are improved in the order of panels 41, 42, 45, and 48 (the light-emitting intensity increases and the chromaticity coordinate y decreases). This shows that the higher a degree the heating temperature in bonding the front panel 10 and back panel 20 is set to, the more the light-emitting characteristics of the PDPs are improved.

This is considered to be because when the panels are preparatively heated to a high temperature while they are separated from each other before they are bonded, a smaller amount of gas is released in the inner space between panels after the panels are bonded since the gas released from the panels is exhausted sufficiently.

By comparing the light-emitting characteristics of the panels 43 to 46 (which have the same temperature profile in the bonding process), it is noted that the light-emitting characteristics are improved in the order of panels 43, 44, 45, and 46 (the chromaticity coordinate y decreases in the order). This shows that the lower the partial pressure of steam vapor in the atmospheric gas is, the more the light-emitting characteristics of the PDPs are improved.

By comparing the light-emitting characteristics of the panels 46 and 47 (which have the same temperature profile in the bonding process), it is noted that the panel 46 is a little superior to the panel 47.

It is considered that this is because a part of oxygen came out of the fluorescent substance being an oxide and the oxygen defect was caused in the panel 47 since it was preparatively heated in the atmosphere of non oxygen, while the panel 46 was preparatively heated in the atmospheric gas containing oxygen.

It is noted that the light-emitting characteristics of the panels 48 and 51 are almost the same. This shows that there is hardly a difference in terms of the light-emitting characteristics of PDPS between a case in which the panels are preparatively heated while they are completely separated from each other and a case in which they are partially separated.

It is noted from Table 5 that the values of the chromaticity coordinate y are almost the same regardless whether they are measured by radiating vacuum ultraviolet rays onto the blue fluorescent substance layer or by emitting light from only the blue fluorescent substance layer.

By focusing attention on the relationships between the chromaticity coordinate y of the emitted blue light and the peak wavelength of the emitted blue light for each panel provided in Table 5, it is noted that the peak wavelength is shorter as the chromaticity coordinate y is smaller. This shows that they are proportional to each other.

<Embodiment 6>

The PDP of the present embodiment has the same

construction as that of Embodiment 1.

The manufacturing method of the PDP is also the same as Embodiment 5 except that after the sealing glass is applied to at least one of the front panel 10 and the back panel 20, the
5 temporary baking process, the bonding process, and the exhausting process are consecutively performed in the heating furnace 81 of the bonding apparatus 80.

The temporary baking process, the bonding process, and the exhausting process of the present embodiment will be
10 described in detail.

These processes are performed using the bonding apparatus shown in FIGs. 19 and 20. However, in the present embodiment, as shown in FIGs. 27A to 27C, a pipe 90 is inserted from outside the heating furnace 81 and connected to the glass
15 pipe 26 which is attached to the air vent 21a of the back panel 20.

FIGs. 27A, 27B, and 27C show operations performed in the temporary baking process through the exhausting process using the bonding apparatus.

20 The temporary baking process, the bonding process, and the exhausting process will be described with reference to these figures.

Temporary Baking Process

A sealing glass paste is applied to one of: the outer
25 region of the front panel 10 on a side facing the back panel 20;

the outer region of the back panel 20 on a side facing the front panel 10; and the outer region of the front panel 10 and the back panel 20 on sides that face each other. Note that in the drawings, the sealing glass layers 15 are formed on the front
5 panel 10.

The front panel 10 and the back panel 20 are put together after positioned properly. The panels are then laid on the base 84 at a fixed position. The pressing mechanisms 86 are then set to press the back panel 20 (FIG. 27A).

10 The atmospheric gas (dry air) is then circulated in the heating furnace 81 (or, at the same time, gas is exhausted through the gas exhaust valve 83 to produce a vacuum) while the following operations are performed.

The slide pins 85 are hoisted to move the back panel
15 20 to a position parallel to itself (FIG. 27B). This broadens the space between the front panel 10 and the back panel 20, and the fluorescent substance layers 25 on the back panel 20 are exposed to the large space in the heating furnace 81.

The heating furnace 81 in the above state is heated
20 to the temporary baking temperature (about 350°C) then the panels are temporarily heated for 10 to 30 minutes at the temperature.

Preparative Heating Process

The panels 10 and 20 are further heated to let the
25 panels release gas having been held by adsorption on the panels.

The preparative heating process ends when a preset temperature (e.g., 400°C) has been reached.

Bonding Process

5 The slide pins 85 are lowered to put the front and back panels together again. That is, the back panel 20 is reset to its proper position on the front panel 10 (FIG. 27C).

10 When the inside of the heating furnace 81 has reached a certain bonding temperature (around 450°C) higher than the softening point of the sealing glass layers 15, the bonding temperature is maintained for 10 to 20 minutes. During this period, the outer regions of the front panel 10 and the back panel 20 are bonded together by the softened sealing glass. Since the back panel 20 is pressed onto the front panel 10 by the pressing mechanisms 86 during this bonding period, the
15 panels are bonded with high stability.

Exhausting Process

The interior of the heating furnace is cooled to an exhaust temperature lower than the softening point of the sealing glass layers 15. The panels are baked at the
20 temperature (e.g., for one hour at 350°C). Gas is exhausted from the inner space between the bonded panels to produce a high degree of vacuum (8×10^{-7} Torr). The exhausting process is performed using a vacuum pump (not illustrated) connected to the pipe 90.

25 The panels are then cooled to room temperature while

the vacuum of the inner space is maintained. The discharge gas is charged into the inner space through the glass pipe 26. The PDP is complete after the air vent 21a is plugged and the glass pipe 26 is cut away.

5 Effects of the Manufacturing Method Shown in the Present Embodiment

The manufacturing method of the present embodiment has the following effects which are not obtained by the conventional methods.

10 Conventionally, the temporary baking process, the bonding process, and the exhausting process are separately performed using a heating furnace, and the panels are cooled to room temperature at each interval between processes. With such a construction, it requires a long time and consumes much energy
15 for the panels to be heated in each process. On the contrary, in the present embodiment, these processes are consecutively performed in the same heating furnace without lowering the temperature to room temperature. This reduces the time and energy required for heating.

20 In the present embodiment, the temporary baking process through the bonding process are performed speedily and with low energy consumption since the temporary baking process and the preparative heating process are performed in the middle of heating the heating furnace 81 to the temperature for the
25 bonding process. Furthermore, in the present embodiment, the

bonding process through the exhausting process are performed speedily and with low energy consumption the exhausting process is performed in the middle of cooling the panels to room temperature after the bonding process.

5 Further, the present embodiment has the same effects as Embodiment 5 compared to conventional bonding methods as will be described.

In general, gases like steam vapor are held by adsorption on the surface of the front panel and back panel.
10 The adsorbed gases are released when the panels are heated.

In conventional methods, in the bonding process after the temporary baking process, the front panel and the back panel are first put together at room temperature, then they are heated to be bonded together. In the bonding process, the gases held
15 by adsorption on the surface of the front panel and back panel are released. Though a certain amount of the gases are released in the temporary baking process, gases are newly held by adsorption when the panels are laid in the air to room temperature before the bonding process begins, and the gases are
20 released in the bonding process. The released gases are confined in the small space between the panels. When this happens, the fluorescent substance layers are tend to be degraded by the heat and the gases, especially by steam vapor released from the protecting layer 14. The degradation of the
25 fluorescent substance layers decreases the light-emitting

intensity of the layers.

On the other hand, according to the manufacturing method shown in the present embodiment, the gas released from the panels are not confined in the inner space since a broad gap
5 is formed between the panels in the bonding process or the preparative heating process. Also, water or the like is not held by adsorption on the panels after the preparative heating process since the panels are consecutively heated in the bonding process following the preparative heating process. Therefore,
10 a small amount of gas is released from the panels during the bonding process. This prevents the fluorescent substance layer 25 from being degraded by heat.

Also, it is possible with the bonding apparatus 80 of the present embodiment to bond the panels at a proper position
15 when the position is properly adjusted at first.

Further, in the present embodiment, the preparative heating process through the bonding process are performed in the atmosphere in which dry gas is circulated. This prevents the fluorescent substance layer 25 from being degraded by heat and
20 the steam vapor contained in the atmospheric gas.

The preferable conditions for the present embodiment in terms of: the temperature in the preparative heating; the timing with which the panels are put together; the type of atmospheric gas; the pressure; and the partial pressure of steam
25 vapor are the same as described in Embodiment 5.

Variations of Present Embodiment

In the present embodiment, the temporary baking process, the preparative heating process, the bonding process, and the exhausting process are consecutively performed in the same apparatus. However, the same effects are obtained to some extent when the preparative heating process is omitted. Also, the same effects are obtained to some extent if only the temporary baking process and the bonding process are consecutively performed in the same apparatus, or if only the bonding process and the exhausting process are consecutively performed in the same apparatus.

In the present embodiment, the interior of the heating furnace is cooled to an exhaust temperature (350°C) lower than the softening point of the sealing glass after the bonding process and gas is exhausted at the temperature. However, it is possible to exhaust gas at a temperature as high as that in the bonding process. In this case, the gas is exhausted sufficiently in a short time. However, to do this, it is thought that some arrangement should be made so that the sealing glass layer does not flow out of the position even if it is softened (e.g., a partition shown in FIGs. 10 to 16).

In the present embodiment, the temporary baking process and the preparative heating process are performed while the front panel 10 and the back panel 20 are separated from each other. However, it is possible to consecutively perform the

temporary baking process, bonding process, and exhausting process adopting the method of Embodiment 3 in which the panels are put together after properly positioned, then the panels are heated to be bonded while the pressure of the inner space is reduced and dry air is supplied to the inner space.

The above method will be detailed. The heating-for-sealing apparatus 50 shown in FIG. 4 is used. First, the sealing glass is applied onto one or both of the front panel 10 and back panel 20 to form the sealing glass layer 15. The panels 10 and 20 are properly positioned then put together without being temporarily baked, and placed in the heating furnace 51.

A pipes 52a is connected to the glass pipes 26a which is attached to the air vent 21a of the back panel 20. Gas is exhausted from the space through the pipe 52b using a vacuum pump (not illustrated). At the same time, dry air is supplied into the inner space through a pipe 52b connected to the glass pipes 26b which is attached to the air vent 21b of the back panel 20. By doing so, the pressure of the inner space is reduced while dry air is flown through the inner space.

With the above state of the space between the panel 10 and 20 maintained, the interior of the heating furnace 51 is heated to a temporary baking temperature and the panels are temporarily baked (for 10 to 30 minutes at 350°C).

Here, the panels are not baked sufficiently in the

temporarily baking if they are simply baked after they are put together since it is difficult for oxygen to be supplied to the sealing glass layer. However, the panels are sufficiently baked if they are baked while dry air is flown through the inner space
5 between the panels.

The temperature is raised to a certain bonding temperature higher than the softening point of the sealing glass and the bonding temperature is maintained for a certain period (e.g., the peak temperature of 450°C is kept for 30 minutes).
10 During this period, the front panel 10 and the back panel 20 are bonded together by the softened sealing glass.

The interior of the heating furnace 51 is cooled to an exhaust temperature lower than the softening point of the sealing glass. Gas is exhausted from the inner space between
15 the bonded panels to produce a high degree of vacuum by maintaining the exhaust temperature. After this exhausting process, the panels are cooled to room temperature. The discharge gas is charged into the inner space through the glass pipe 26. The PDP is complete after the air vent 21a is plugged
20 and the glass pipe 26 is cut away.

In this variation example, as in the method of the present embodiment, the temporary baking, bonding, and exhausting processes are consecutively performed in the same bonding apparatus while the temperature does not decrease to
25 room temperature. Therefore, these process are also performed

speedily and with low energy consumption.

In this variation example, the same effects are obtained to some extent if only the temporary baking process and the bonding process are consecutively performed in the heating
5 furnace 51, or if only the bonding process and the exhausting process are consecutively performed in the heating furnace 51.

Example 6

<Table 6>

TABLE 6.

PANEL No.	TEMPERATURE FOR TEMPORARILY BAKING FRIT(°C)	TEMPERATURE FOR PUTTING FRONT AND BACK PANELS TOGETHER(°C)	TEMPERATURE FOR BONDING PANELS(°C)	TEMPERATURE FOR EXHAUSTING GAS(°C)	ATMOSPHERE DURING BONDING	PARTIAL PRESSURE OF STEAM VAPOR IN DRY AIR (Torr)	RELATIVE LIGHT-EMITTING INTENSITY OF BLUE LIGHT	CHROMATICITY COORDINATE Y OF BLUE LIGHT	COLOR TEMPERATURE IN WHITE BALANCE (K)	PEAK INTENSITY RATIO OF SPECTRUM OF BLUE AND GREEN LIGHT (BLUE/GREEN)	PEAK NUMBER OF MOLECULES IN H ₂ O GAS DESORBED FROM BLUE FLUORESCENT SUBSTANCE AT 200 °C OR MORE WITH TDS ANALYSIS	AXIS LENGTH RATIO OF BLUE FLUORESCENT SUBSTANCE CRYSTAL (c-AXIS/a-AXIS)
61	350	250	450	350	DRY AIR	2	107	0.078	6700	0.80	1.0×10^{16}	4.02180
62	350	350	450	350	DRY AIR	2	118	0.057	8600	0.95	4.0×10^{15}	4.02172
63	350	400	450	350	DRY AIR	12	108	0.075	7100	0.82	7.3×10^{15}	4.02178
64	350	400	450	350	DRY AIR	8	112	0.065	7800	0.91	5.2×10^{15}	4.02174
65	350	400	450	350	DRY AIR	2	120	0.055	9000	0.98	3.4×10^{15}	4.02168
66	350	400	450	350	DRY AIR	0	123	0.053	9800	1.09	2.3×10^{15}	4.02165
67	350	400	450	350	VACUUM	-	120	0.053	9300	1.03	1.3×10^{15}	4.02155
68	350	450	450	350	DRY AIR	2	125	0.052	10600	1.15	1.9×10^{15}	4.02160
69	350	480	450	350	DRY AIR	2	126	0.052	11000	1.19	1.3×10^{15}	4.02155
70	350	-	450	350	AIR	-	100	0.090	5800	0.67	2.6×10^{16}	4.02208

The panels 61 to 69 are PDPs manufactured based on the present embodiment. The panels 61 to 69 have been manufactured in different conditions during the bonding process. That is, the panels were heated in various types of atmospheric gases under various pressures, and they were put together at various temperatures with various timing.

FIG. 28 shows the temperature profile used in the temporary baking process, bonding process, and exhausting process in manufacturing the panels 63 to 67.

For the panels 61 to 66, 68, and 69, dry air with different partial pressures of steam vapor in the range of 0Torr to 12Torr were used. For panel 70, non-dry air was used. The panel 67 was heated while gas was exhausted to produce a vacuum.

For the panels 63 to 67, the panels were heated from the room temperature to 350°C. The panels were temporarily baked by maintaining the temperature for 10 minutes. The panels were then heated to 400°C (lower than the softening point of sealing glass), then the panels were put together. The panels were further heated to 450°C (higher than the softening point of sealing glass), the temperature was maintained for 10 minutes then decreased to 350°C, and gas was exhausted while the temperature of 350°C was maintained.

For the panels 61 and 62, the panels were bonded at lower temperatures of 250°C and 350°C, respectively.

For the panel 68, the panels were heated to 450°C, then put together at the temperature. For the panel 69, the panels were heated to the peak temperature of 480°C then decreased to 450°C, and the panels were put together and bonded
5 at 450°C.

The panel 70 is a comparative PDP manufactured based on a conventional method in which the panels were temporarily baked, put together at room temperature, heated to a bonding temperature of 450°C in air at the atmospheric pressure, and
10 bonded at 450°C. The panels were then cooled to room temperature once, then heated again in the heating furnace to an exhaust temperature of 350°C. Gas was exhausted from the space by maintaining the temperature at 350°C.

Note that in each of the PDPs 61 to 70, the thickness
15 of the fluorescent substance layer is 30μm, and the discharge gas, Ne(95%)-Xe(5%), was charged with the charging pressure 500Torr so that each has the same panel construction.

Test for Light Emitting Characteristics

For each of PDPs 61 to 70, the relative light-emitting
20 intensity of the emitted blue light, the chromaticity coordinate y of the emitted blue light, the peak wavelength of the emitted blue light, the color temperature in the white balance without color correction, and the ratio of the peak intensity of the spectrum of light emitted from the blue cells to that of the
25 green cells were measured as the light emitting

characteristics.

The results are shown in Table 6. Note that the relative light-emitting intensity values for blue light shown in Table 6 are relative values when the measured light-emitting intensity of the panel 70, a comparative example, is set to 100 as the standard value.

Each of the manufactured PDPs was disassembled and vacuum ultraviolet rays were radiated onto the blue fluorescent substance layers of the back panel using a krypton excimer lamp. The chromaticity coordinate y of the emitted blue light, the color temperature when light was emitted from all of the blue, red, and green cells, and the ratio of the peak intensity of the spectrum of light emitted from the blue cells to that of the green cells were then measured. The results were the same as the above ones.

The blue fluorescent substances were then taken out from the panel. The number of molecules contained in one gram of H_2O gas desorbed from the blue fluorescent substances was measured using the TDS analysis method. Also, the ratio of c -axis length to a -axis length of the blue fluorescent substance crystal was measured by the X-ray analysis. The results are also shown in Table 6.

Study

For each of the PDPs 61 to 70, the light-emitting intensity of the emitted blue light, the chromaticity coordinate

y of the emitted blue light, the peak wavelength of the emitted blue light, and the color temperature in the white balance without color correction (a color temperature when light is emitted from the blue, red, and green cells with the same power to produce a white display) were measured as the light emitting characteristics.

<Test Results>

The results of this test are shown in Table 6. Note that the relative light-emitting intensity values for blue light shown in Table 6 are relative values when the measured light-emitting intensity of the panel 70 is set to 100 as the standard value.

It is noted from the Table 6 that the panels 61 to 69 have light emitting characteristics superior to those of the panel 70 (with higher light-emitting intensity of blue light and smaller chromaticity coordinate y). It is thought that this is because a smaller amount of gas is released in the inner space between panels after the panels are bonded in accordance with the present embodiment than in accordance with conventional methods.

In the PDP of panel 70, the chromaticity coordinate y of the light emitted from blue cells is 0.090 and the color temperature in the white balance without color correction is 5800K. In contrast, in panels 61 to 69, the values are respectively 0.08 or less and 6500K or more. Especially, it is

noted that in panels 68 and 69 that have low chromaticity coordinate y of blue light, a high color temperature of around 11,000K has been achieved (in the white balance without color correction).

5 By comparing the light-emitting characteristics of the panels 61, 62, 65, 68, and 69 (in each of which the partial pressure of steam vapor in the dry gas is 2Torr), it is noted that the light-emitting characteristics are improved in the order of panels 61, 62, 65, 68, 69 (the light-emitting intensity increases and the chromaticity coordinate y decreases). This shows that the higher a degree the heating temperature in bonding the front panel 10 and back panel 20 is set to, the more the light-emitting characteristics of the PDPs are improved.

15 By comparing the light-emitting characteristics of the panels 63 to 66 (which have the same temperature profile in the bonding process), it is noted that the light-emitting characteristics are improved in the order of panels 63, 64, 65, and 66 (the chromaticity coordinate y decreases in the order). This shows that the lower the partial pressure of steam vapor in the atmospheric gas is, the more the light-emitting characteristics of the PDPs are improved.

20 By comparing the light-emitting characteristics of the panels 66 and 67 (which have the same temperature profile in the bonding process), it is noted that the panel 66 is a little superior to the panel 67.

It is considered that this is because a part of oxygen came out of the fluorescent substance being an oxide and the oxygen defect was caused in the panel 67 since it was preparatively heated in the atmosphere of non oxygen, while the panel 66 was preparatively heated in the atmospheric gas containing oxygen.

Others

in the above Embodiments 1 to 6, the case of manufacturing a surface-discharge type PDP was described. However, the present invention can be applied to the case of manufacturing an opposed-discharge type PDP.

The present invention can be achieved by using the fluorescent substances generally used for PDPs other than the fluorescent substances with the composition shown in the above embodiments.

Typically, the sealing glass is applied after the the fluorescent substance layer is formed, as shown in Embodiments 1 to 6. However, the order of these process may be reversed.

INDUSTRIAL USE POSSIBILITY

The PDP of the present invention and the method of producing the PDP are effective for manufacturing displays for computers or TVs, especially for manufacturing large-screen displays.